Classify each of the materials below. In the center column, state whether the material is a **pure** substance or a mixture.

- If the material is a <u>pure substance</u>, further classify it as either an **element** or **compound** in the right column.
- If the material is a mixture, further classify it as **homogeneous** or **heterogeneous** in the right column. Write the entire word in each space to earn full credit.

Material	Pure Substance _	→ Element or Compound
	Mixture ——	Solution(Homogeneous) or Mechanical mixture (Heterogeneous)
Laundry detergent (contains white and blue crystals)		
sugar + pure water $(C_{12}H_{22}O_{11} + H_2O)$		
iron filings (Fe)		
limestone (CaCO ₃)		
orange juice (water and pulp)	:	
Pacific Ocean (Water and Salt)		
air		
aluminum (AI)		
magnesium (Mg)		
acetylene (C ₂ H ₂)		
tap water in a glass		
pure water (H ₂ O)		
soil		
chromium (Cr)		
baking soda (NaHCO ₃)		
salt + pure water (NaCl + H ₂ O)		
benzene (C ₆ H ₆)		
muddy water		
brass (Cu mixed with Zn)		
Pizza		

Classify the following statements as heterogeneous mixture, a suspension, a colloid or a solution (homogeneous mixture).

- 1. Spicy salad dressing
- 2. Soda pop
- 3. Fertilizer that is spread on the lawn.
- 4. Paint for the walls
- 5. A compost pile in the backyard for the garden
- 6. A gold necklace (gold and copper)
- 7. Laundry detergent (contains white and blue crystals)
- 8. Pizza
- 9. Cough syrup that must be shaken before it is ingested.
- 10. Freshly squeezed orange juice
- 11. Blueberry muffin
- 12. Benzene (C₆H₆)
- 13. Nail polish remover (C₃H₆O)
- 14. A muddy pond
- 15. A mixture of rocks, grass and metal