Unit 4: Periodic Trends Periodic Trends Jigsaw Activity

Name	
Date	Block

Periodic Trend Research

1.	Which trend are you researching? Atomic Radius Trend
2.	How is atomic radius measured?
3.	Define "nuclear force".
	Define "electromagnetic force".
5.	How does the electromagnetic force affect the size of an atom?
6.	Define "electron - electron repulsion".
7.	How does electron - electron repulsion affect the size of an atom?
8.	How does Coulomb's Law affect this trend?
9.	What happens to the trend moving from left to right across a period?
Э.	What happens to the trend moving down a group?
1.	Explain how the trend is different for metals versus nonmetals.
2.	Which group/ element has the largest value for this trend?//
3.	Which group/ element has the smallest value for this trend?//
4.	How do you think this trend affects chemical reactions ?
5.	How are other Periodic trends related to atomic radius?