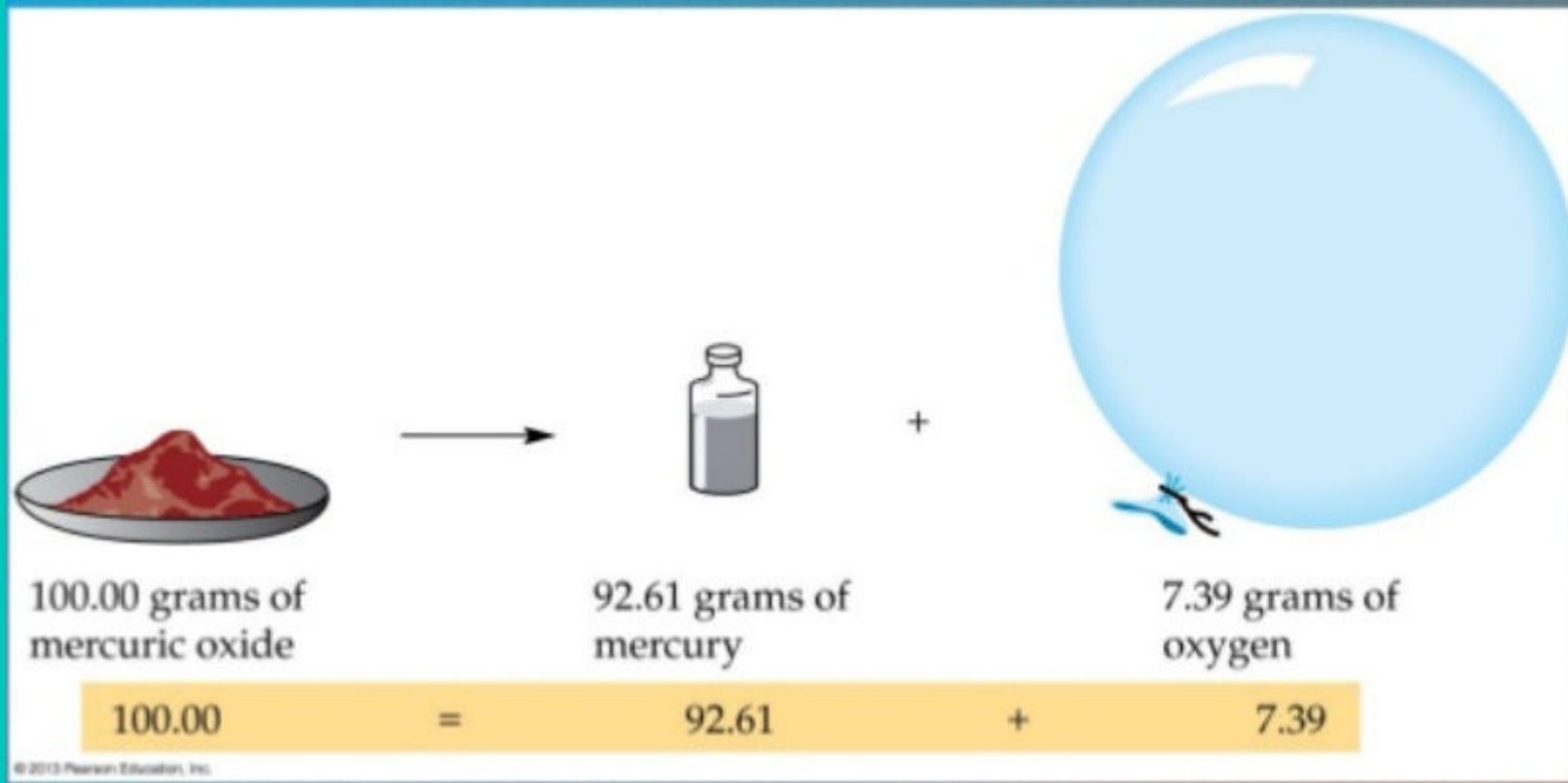
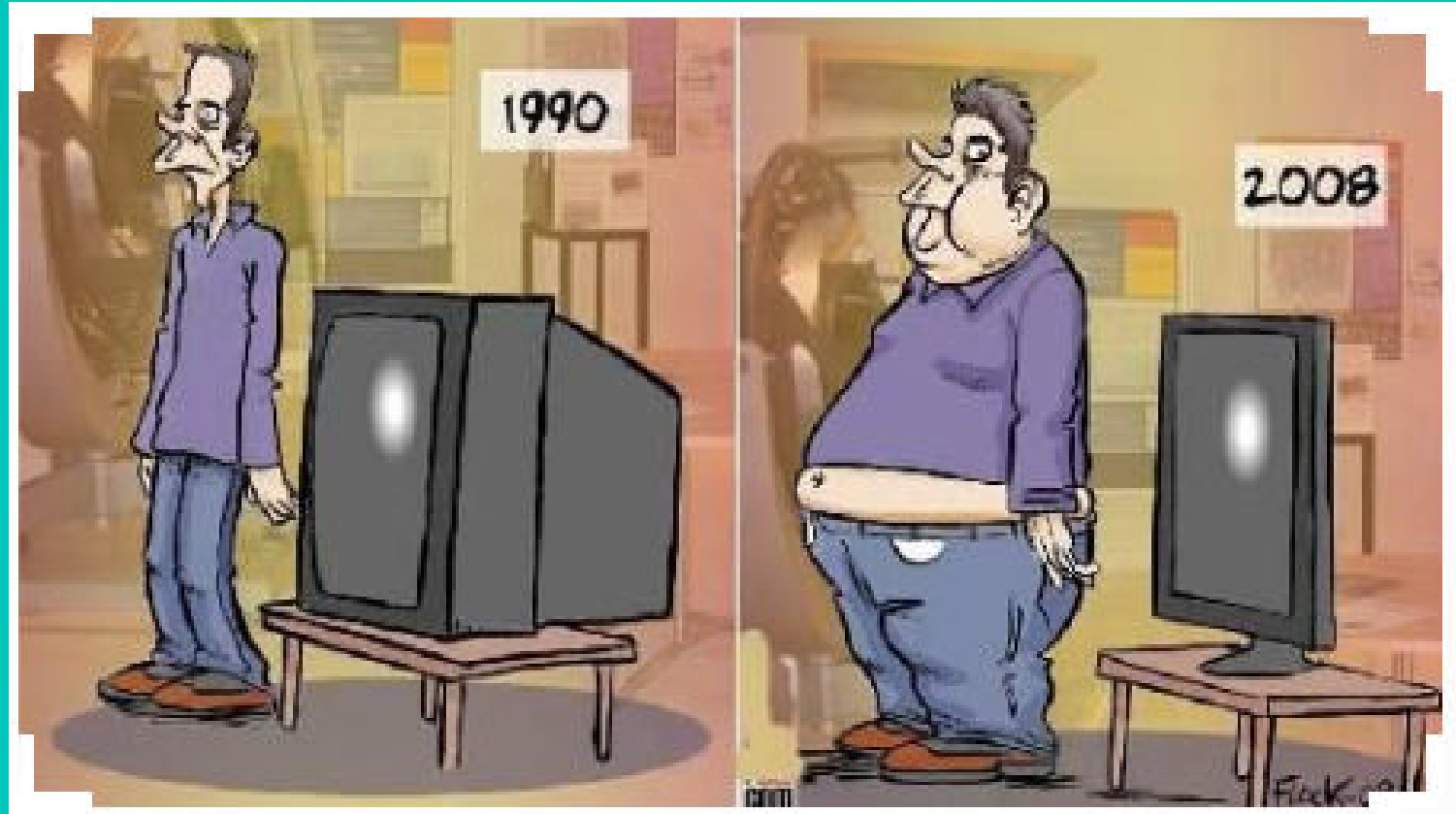


Laws of chemical composition



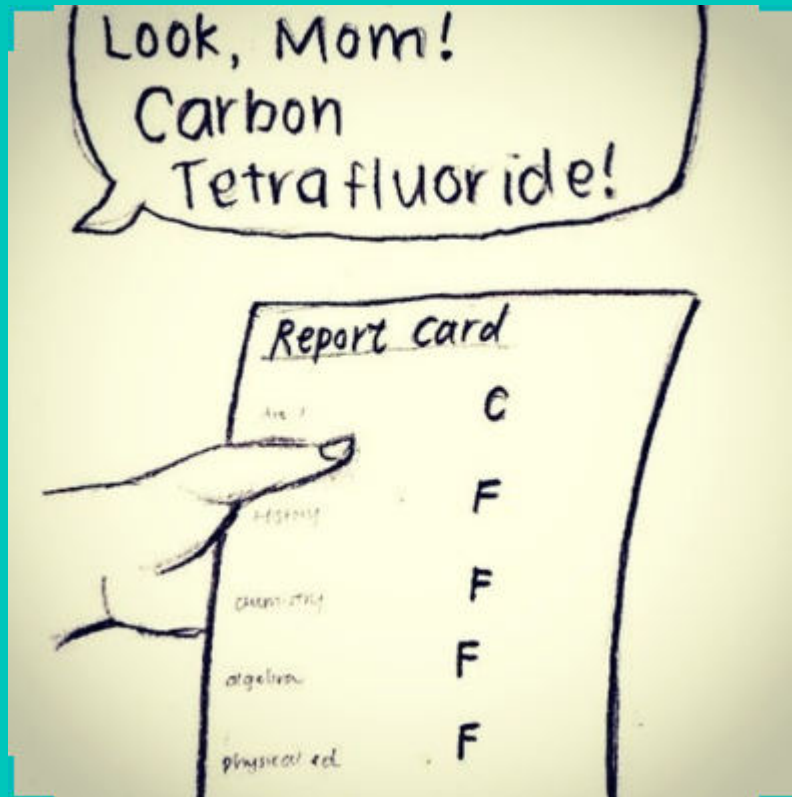
Law of Mass Conservation






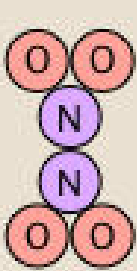
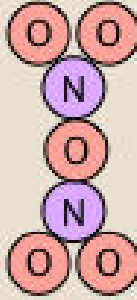
Law of Mass Conservation



Law of Definite Proportions

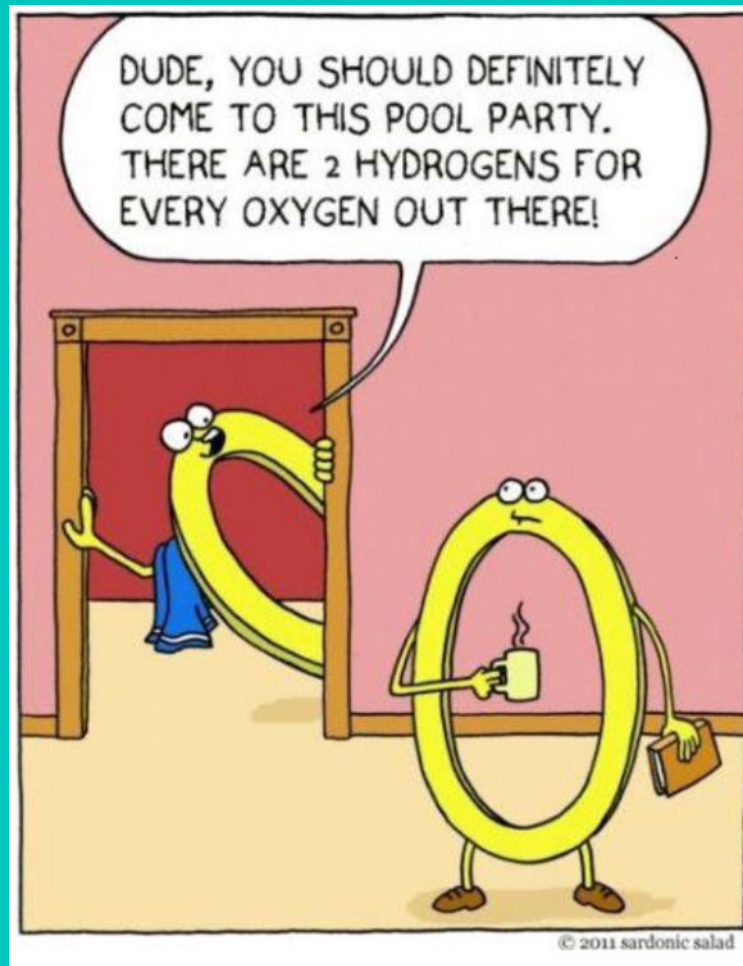


Law of Definite Proportions

					
	NO	NO ₂	N ₂ O	N ₂ O ₄	N ₂ O ₅
1	14:16	14:32	28:16	28:48	28:80
2	1.14	2.29	0.571	1.71	2.86
3	2	4	1	3	5
	ratio of molar masses N:O				
	grams of O combining with 1 g of N				
	divide through by smallest O:N mass ratio (.571)				

Stephen Lower

Law of Multiple Proportions



Law of Multiple Proportions