

block 7- atomic radius trend

Played on	8 Nov 2019
Hosted by	sophie.milani
Played with	24 players
Played	10 of 10

Overall Performance

Total correct answers (%)	65,00%
Total incorrect answers (%)	35,00%
Average score (points)	6667,1



Feedback

Number of responses	0
How fun was it? (out of 5)	0,00 o
Did you learn something?	0,00%
Do you recommend it?	0,00%
How do you feel?	

Switch tabs/pages to view other result breakdown

Overview

%
%
17 points

ut of 5			
Yes	0,00% No		
Yes	0,00% No		
0,00% Positive		0,00% Neutral	

--

Overview

Response Category	Percentage
Very much	~15%
Quite much	~35%
Somewhat	~35%
Not at all	~15%

block 7- atomic radius trend

Final Scores

Rank	Players
1	Ryan
2	sophie
3	Daniel
4	lauren
5	Julian
6	Willow
7	Henry
8	niklas
9	victoria
10	Dana
11	kevy
12	Orion
13	Will C
14	balin
15	Garrett Keeney
16	Brandon
17	Bianca
18	fernanda
19	Kent
20	♠William L♠
21	carly
22	Savannah

Final Scores

23	Shane
24	Emily

Final Scores

Total Score (points)	Correct Answers	Incorrect Answers
9655	8	2
9645	8	2
9547	8	2
9203	8	2
9007	8	2
8837	8	2
8298	8	2
7950	7	3
7250	8	2
7202	7	3
6329	6	4
6240	7	3
6239	7	3
6157	6	4
5705	6	4
5701	6	4
5480	6	4
4896	5	5
4771	5	5
4730	5	5
4678	5	5
4615	5	5

Final Scores

4425	5	5
3452	4	6

block 7- atomic radius trend

Kahoot! Summary

Rank	Players
1	Ryan
2	sophie
3	Daniel
4	lauren
5	Julian
6	Willow
7	Henry
8	niklas
9	victoria
10	Dana
11	kevy
12	Orion
13	Will C
14	balin
15	Garrett Keeney

Kahoot! Summary

16	Brandon
17	Bianca
18	fernanda
19	Kent
20	♠William L♠
21	carly
22	Savannah
23	Shane
24	Emily

Kahoot! Summary

Total Score (points)	Q1
9655	928
9645	962
9547	897
9203	843
9007	955
8837	765
8298	663
7950	958
7250	873
7202	953
6329	933
6240	940
6239	865
6157	932
5705	905

Kahoot! Summary

5701	860
5480	838
4896	957
4771	940
4730	970
4678	942
4615	922
4425	803
3452	960

Kahoot! Summary

what is the radius?	Q2
half the diameter of a circle	925
half the diameter of a circle	890
half the diameter of a circle	945
half the diameter of a circle	990
half the diameter of a circle	893
half the diameter of a circle	947
half the diameter of a circle	735
half the diameter of a circle	0
half the diameter of a circle	908
half the diameter of a circle	897
half the diameter of a circle	1027
half the diameter of a circle	858
half the diameter of a circle	917
half the diameter of a circle	1018
half the diameter of a circle	0

Kahoot! Summary

half the diameter of a circle	883
half the diameter of a circle	732
half the diameter of a circle	1007
half the diameter of a circle	0
half the diameter of a circle	0
half the diameter of a circle	988
half the diameter of a circle	0
half the diameter of a circle	935
half the diameter of a circle	0

Kahoot! Summary

How is the atomic radius measures?	Q3
Half of the distance between the nuclei of two atoms	1177
Half of the distance between the nuclei of two atoms	1163
Half of the distance between the nuclei of two atoms	1147
Half of the distance between the nuclei of two atoms	1150
Half of the distance between the nuclei of two atoms	1105
Half of the distance between the nuclei of two atoms	940
Half of the distance between the nuclei of two atoms	1050
the distance from the center of a nucleus to the outside of the atom	983
Half of the distance between the nuclei of two atoms	1138
Half of the distance between the nuclei of two atoms	1177
Half of the distance between the nuclei of two atoms	1170
Half of the distance between the nuclei of two atoms	963
Half of the distance between the nuclei of two atoms	967
Half of the distance between the nuclei of two atoms	1182
the distance from the outside of one atom to the outside of another	955

Kahoot! Summary

Half of the distance between the nuclei of two atoms	1133
Half of the distance between the nuclei of two atoms	1085
Half of the distance between the nuclei of two atoms	1072
the distance from the center of a nucleus to the outside of the atom	972
the distance from the outside of one atom to the outside of another	980
Half of the distance between the nuclei of two atoms	1063
the distance from the outside of one atom to the outside of another	973
Half of the distance between the nuclei of two atoms	1175
the distance from the center of a nucleus to the outside of the atom	862

Kahoot! Summary

The nuclear force is the force between nucleons in the nucleus that holds the nucleus together	Q4	
True	1185	
True	1258	
True	1218	
True	1255	
True	1242	
True	1207	
True	895	
True	1062	
True	0	
True	1248	
True	1285	
True	0	
True	0	
True	1278	
True	0	

Kahoot! Summary

	True	0
	True	1135
	True	0
	True	1073
	True	0
	True	0
	True	0
	True	0
	True	0

Kahoot! Summary

The nuclear force holds protons together because opposite charges attract		Q5
	False	1283
	False	1242
	False	1243
	False	1082
	False	1207
	False	945
	False	1067
	False	1045
	True	545
	False	0
	False	0
	True	568
	True	590
	False	0
	True	773

Kahoot! Summary

	True	0
	False	0
	True	0
	False	0
	True	0
	True	0
	True	0
	True	0
	True	0

Kahoot! Summary

Through nuclear force, which atom is the largest size?		Q6
	Na	1397
	Na	1360
	Na	1348
	Na	1278
	Na	1348
	Na	1287
	Na	1293
	Na	1212
	Na	663
	Mg	830
	Ar	0
	Na	885
	Na	905
	Mg	0
	Na	867

Kahoot! Summary

	Ar	770
	Ar	0
	Mg	812
	P	0
	Ar	910
	Mg	800
	Ar	875
	Ar	520
	Ar	802

Kahoot! Summary

What's the principle of electron-electron repulsion?	Q7
The principle that electron pairs orient themselves as far as possible	1360
The principle that electron pairs orient themselves as far as possible	1375
The principle that electron pairs orient themselves as far as possible	1387
The principle that electron pairs orient themselves as far as possible	1293
The principle that electron pairs orient themselves as far as possible	1362
The principle that electron pairs orient themselves as far as possible	1363
The principle that electron pairs orient themselves as far as possible	1302
The principle that electron pairs orient themselves as far as possible	1322
The principle that electron pairs orient themselves as far as possible	1115
The principle that electron pairs orient themselves as far as possible	982
Principle that electrons come together	917
The principle that electron pairs orient themselves as far as possible	1048
The principle that electron pairs orient themselves as far as possible	1022
Principle where protons are repelled from each other	797
The principle that electron pairs orient themselves as far as possible	1068

Kahoot! Summary

The principle that electron pairs orient themselves as far as possible	920
Principle where protons are repelled from each other	867
The principle that electron pairs orient themselves as far as possible	1048
Principle where protons are repelled from each other	813
The principle that electron pairs orient themselves as far as possible	978
The principle that electron pairs orient themselves as far as possible	885
The principle that electron pairs orient themselves as far as possible	947
The principle that electron pairs orient themselves as far as possible	992
The principle that electron pairs orient themselves as far as possible	828

Kahoot! Summary

How does electron-electron repulsion affect the size of an atom?	Q8
It makes it larger	1400
It makes it larger	1395
It makes it larger	1362
It makes it larger	1312
It makes it larger	0
It makes it larger	1383
It makes it larger	1293
It makes it larger	1368
It makes it larger	1105
It makes it larger	1115
It makes it larger	997
It makes it larger	978
It makes it larger	973
It makes it larger	950
It makes it larger	1137

Kahoot! Summary

It makes it larger	1135
It makes it larger	823
It makes it larger	0
It makes it larger	973
It makes it larger	892
It makes it larger	0
It makes it larger	898
It makes it larger	0
It makes it larger	0

Kahoot! Summary

How does coulomb's law affect the atomic radius trend?	Q9
The attraction gets stronger	0
The attraction gets stronger	0
The attraction gets stronger	0
The attraction gets stronger	0
the atomic radius doubles	0
The attraction gets stronger	0
The attraction gets stronger	0
The attraction gets stronger	0
The attraction gets stronger	0
The attraction gets stronger	0
The attraction gets stronger	0
The attraction gets stronger	0
The attraction gets stronger	0
The attraction gets stronger	0
The attraction gets stronger	0

Kahoot! Summary

The attraction gets stronger	0
The attraction gets stronger	0
The attraction gets weaker	0
The attraction gets stronger	0
The attraction gets stronger	0
The attraction gets weaker	0
The attraction gets stronger	0
The attraction gets weaker	0
The attraction does nothing	0

Kahoot! Summary

What happens to the trend when moving left to right across a period?	Q10
Electronegativity from left to right decreases	0
Electronegativity from left to right decreases	0
Electronegativity from left to right decreases	0
Electronegativity from left to right decreases	0
Electronegativity from left to right decreases	895
	0
Electronegativity from left to right decreases	0
Electronegativity from left to right decreases	0
	903
	0
Electronegativity from left to right decreases	0
Electronegativity from left to right decreases	0
Electronegativity from left to right decreases	0
Electronegativity from left to right decreases	0
Electronegativity from left to right decreases	0

Kahoot! Summary

	0
	0
Electronegativity from left to right decreases	0
	0
Electronegativity from left to right decreases	0
Electronegativity from left to right decreases	0
Electronegativity from left to right decreases	0
Electronegativity from left to right decreases	0
	0

What happens to a trend moving down a group?
Electronegativity of an element increases
Electronegativity of an element increases
Electronegativity of an element increases
Electronegativity of an element decreases
Electronegativity of an element increases
Electronegativity of an element decreases
Electronegativity of an element increases
Electronegativity of an element increases
Electronegativity of an element increases

Kahoot! Summary

Electronegativity of an element increases
Electronegativity of an element increases
Electronegativity of an element increases
Electronegativity of an element increases
Electronegativity of an element increases
Electronegativity of an element increases

block 7- a
1 Quiz
Correct answers
Players correct (
Question duratic
Answer Sum
Answer options
Is answer correc
Number of answ
Average time tal
Answer Deta
Players
Bianca
Brandon
Dana
Daniel
Emily
Garrett Keeney
Henry
Julian
Kent
Orion

1 Quiz

Ryan
Savannah
Shane
Will C
Willow
balin
carly
fernanda
kevy
lauren
niklas
sophie
victoria
♠William L♠

Atomic radius trend	
what is the radius?	
s	half the c
(%)	100,00%
on	30 secur

Summary	
	▲
st?	
ers received	
ken to answer (seconds)	


ails	
	Answer
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓

1 Quiz

	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓

1 Quiz

diameter of a circle
,
nds

the distance around the circle	
X	
0	
0,00	

	Score (p
half the diameter of a circle	838
half the diameter of a circle	860
half the diameter of a circle	953
half the diameter of a circle	897
half the diameter of a circle	960
half the diameter of a circle	905
half the diameter of a circle	663
half the diameter of a circle	955
half the diameter of a circle	940
half the diameter of a circle	940

1 Quiz

half the diameter of a circle	928
half the diameter of a circle	922
half the diameter of a circle	803
half the diameter of a circle	865
half the diameter of a circle	765
half the diameter of a circle	932
half the diameter of a circle	942
half the diameter of a circle	957
half the diameter of a circle	933
half the diameter of a circle	843
half the diameter of a circle	958
half the diameter of a circle	962
half the diameter of a circle	873
half the diameter of a circle	970

1 Quiz

half the diameter of a circle	<div></div>
<div>✓</div>	
24	
6,09	

oints)	Current
	838
	860
	953
	897
	960
	905
	663
	955
	940
	940

1 Quiz

	928
	922
	803
	865
	765
	932
	942
	957
	933
	843
	958
	962
	873
	970

1 Quiz

the distance from one circle to another	<input type="checkbox"/>
X	
0	
0,00	

Total Score (points)	Answer ti
	9,7
	8,4
	2,8
	6,2
	2,4
	5,7
	20,2
	2,7
	3,6
	3,6

1 Quiz

	4,3
	4,7
	11,8
	8,1
	14,1
	4,1
	3,5
	2,6
	4
	9,4
	2,5
	2,3
	7,6
	1,8

1 Quiz

block 7- a
2 Quiz
Correct answers
Players correct (
Question duratic
Answer Sum
Answer options
Is answer correc
Number of answ
Average time tal
Answer Deta
Players
Bianca
Brandon
Dana
Daniel
Emily
Garrett Keeney
Henry
Julian
Kent
Orion

2 Quiz

Ryan
Savannah
Shane
Will C
Willow
balin
carly
fernanda
kevy
lauren
niklas
sophie
victoria
♠William L♠

Atomic radius trend	
How is the atomic radius measures?	
s	Half of th
(%)	75,00%
on	30 secur

Summary	
	▲
st?	
ers received	
ken to answer (seconds)	

ails	
	Answer
	✓
	✓
	✓
	✓
	✗
	✗
	✓
	✓
	✗
	✓

2 Quiz

	✓
	✗
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✗
	✓
	✓
	✗

2 Quiz

the distance between the nuclei of two atoms
nds


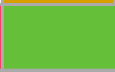
the distance from the center of a nucleus to the outside of the atom	◆
x	
3	
5,80	

	Score (p
Half of the distance between the nuclei of two atoms	732
Half of the distance between the nuclei of two atoms	883
Half of the distance between the nuclei of two atoms	897
Half of the distance between the nuclei of two atoms	945
the distance from the center of a nucleus to the outside of the atom	0
the distance from the outside of one atom to the outside of another	0
Half of the distance between the nuclei of two atoms	735
Half of the distance between the nuclei of two atoms	893
the distance from the center of a nucleus to the outside of the atom	0
Half of the distance between the nuclei of two atoms	858

2 Quiz

Half of the distance between the nuclei of two atoms	925
the distance from the outside of one atom to the outside of another	0
Half of the distance between the nuclei of two atoms	935
Half of the distance between the nuclei of two atoms	917
Half of the distance between the nuclei of two atoms	947
Half of the distance between the nuclei of two atoms	1018
Half of the distance between the nuclei of two atoms	988
Half of the distance between the nuclei of two atoms	1007
Half of the distance between the nuclei of two atoms	1027
Half of the distance between the nuclei of two atoms	990
the distance from the center of a nucleus to the outside of the atom	0
Half of the distance between the nuclei of two atoms	890
Half of the distance between the nuclei of two atoms	908
the distance from the outside of one atom to the outside of another	0

2 Quiz

the distance from the outside of one atom to the outside of another	
X	
3	
9,07	

oints)	Current
	1570
	1743
	1850
	1842
	960
	905
	1398
	1848
	940
	1798

2 Quiz

	1853
	922
	1738
	1782
	1712
	1950
	1930
	1964
	1960
	1833
	958
	1852
	1781
	970

2 Quiz

Half of the distance between the nuclei of two atoms	<input type="checkbox"/>
<input checked="" type="checkbox"/>	
18	
11,02	

Total Score (points)	Answer to the question
	22,1
	13
	12,2
	9,3
	6,2
	12
	21,9
	12,4
	5,5
	14,5

2 Quiz

	10,5
	9,4
	9,9
	11
	9,2
	4,9
	6,7
	5,6
	4,4
	6,6
	5,7
	12,6
	11,5
	5,8

2 Quiz

block 7- a
3 Quiz
Correct answers
Players correct (
Question duratic
Answer Sum
Answer options
Is answer correc
Number of answ
Average time tal
Answer Deta
Players
Bianca
Brandon
Dana
Daniel
Emily
Garrett Keeney
Henry
Julian
Kent
Orion

3 Quiz

Ryan
Savannah
Shane
Will C
Willow
balin
carly
fernanda
kevy
lauren
niklas
sophie
victoria
♠William L♠

Atomic radius trend	
The nuclear force is the force between nucleons in the nucleus that	
s	True
(%)	100,00%
on	30 seconds



Summary	
	▲
st?	
ers received	
ken to answer (seconds)	

Details	
	Answer
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓

3 Quiz

	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓

at holds the nucleus together
,
nds

False	
X	
0	
0,00	

	Score (p
True	1085
True	1133
True	1177
True	1147
True	862
True	955
True	1050
True	1105
True	972
True	963

3 Quiz

True	1177
True	973
True	1175
True	967
True	940
True	1182
True	1063
True	1072
True	1170
True	1150
True	983
True	1163
True	1138
True	980

3 Quiz

True	<div></div>
<div>✔</div>	
24	
5,05	

oints)	Current
	2655
	2876
	3027
	2989
	1822
	1860
	2448
	2953
	1912
	2761

3 Quiz

	3030
	1895
	2913
	2749
	2652
	3132
	2993
	3036
	3130
	2983
	1941
	3015
	2919
	1950

3 Quiz

Total Score (points)	Answer ti
	6,9
	4
	1,4
	3,2
	8,3
	2,7
	9
	5,7
	1,7
	14,2

3 Quiz

	1,4
	1,6
	1,5
	14
	15,6
	1,1
	8,2
	7,7
	1,8
	3
	1
	2,2
	3,7
	1,2

3 Quiz

[illegible]

3 Quiz

block 7- a
4 Quiz
Correct answers
Players correct (
Question duratic
Answer Sum
Answer options
Is answer correc
Number of answ
Average time tal
Answer Deta
Players
Bianca
Brandon
Dana
Daniel
Emily
Garrett Keeney
Henry
Julian
Kent
Orion

4 Quiz

Ryan
Savannah
Shane
Will C
Willow
balin
carly
fernanda
kevy
lauren
niklas
sophie
victoria
♠William L♠

Atomic radius trend	
The nuclear force holds protons together because opposite charge	
s	False
(%)	54,17%
on	30 seconds




Summary	
	▲
st?	
ers received	
ken to answer (seconds)	

Details	
	Answer
	✓
	✗
	✓
	✓
	✗
	✗
	✓
	✓
	✓
	✗

4 Quiz

	✓
	✗
	✗
	✗
	✓
	✓
	✗
	✗
	✓
	✓
	✓
	✓
	✗
	✗

ges attract
nds

False	
	
13	
5,35	

	Score (p
False	1135
True	0
False	1248
False	1218
True	0
True	0
False	895
False	1242
False	1073
True	0

4 Quiz

False	1185
True	0
True	0
True	0
False	1207
False	1278
True	0
True	0
False	1285
False	1255
False	1062
False	1258
True	0
True	0

4 Quiz

True	<input checked="" type="radio"/>
X	
11	
9,45	

oints)	Current
	3790
	2876
	4275
	4207
	1822
	1860
	3343
	4195
	2985
	2761

4 Quiz

	4215
	1895
	2913
	2749
	3859
	4410
	2993
	3036
	4415
	4238
	3003
	4273
	2919
	1950

4 Quiz

Total Score (points)	Answer ti
	9,9
	2,2
	3,1
	4,9
	4,9
	27,5
	24,3
	3,5
	1,6
	14,5

4 Quiz

	6,9
	1,3
	3,5
	15,7
	5,6
	1,3
	6,6
	7,8
	0,9
	2,7
	2,3
	2,5
	18,5
	1,4

4 Quiz

[illegible]

4 Quiz

block 7- a
5 Quiz
Correct answers
Players correct (
Question duratic
Answer Sum
Answer options
Is answer correc
Number of answ
Average time tal
Answer Deta
Players
Bianca
Brandon
Dana
Daniel
Emily
Garrett Keeney
Henry
Julian
Kent
Orion

5 Quiz

Ryan
Savannah
Shane
Will C
Willow
balin
carly
fernanda
kevy
lauren
niklas
sophie
victoria
♠William L♠

Atomic radius trend	
Through nuclear force, which atom is the largest size?	
s	Na
(%)	50,00%
on	30 secur

Summary	
	▲
st?	
ers received	
ken to answer (seconds)	

ails	
	Answer
	X
	X
	X
	✓
	X
	✓
	✓
	✓
	X
	✓

5 Quiz

	✓
	✗
	✗
	✓
	✓
	✗
	✗
	✗
	✗
	✓
	✓
	✓
	✓
	✗

5 Quiz

nds

P	◆
X	
1	
4,30	

	Score (p
Ar	0
Ar	0
Mg	0
Na	1243
Ar	0
Na	773
Na	1067
Na	1207
P	0
Na	568

5 Quiz

Na	1283
Ar	0
Ar	0
Na	590
Na	945
Mg	0
Mg	0
Mg	0
Ar	0
Na	1082
Na	1045
Na	1242
Na	545
Ar	0

5 Quiz

Mg	<div></div>
X	
4	
14,40	

oints)	Current
	3790
	2876
	4275
	5450
	1822
	2633
	4410
	5402
	2985
	3329

5 Quiz

	5498
	1895
	2913
	3339
	4804
	4410
	2993
	3036
	4415
	5320
	4048
	5515
	3464
	1950

5 Quiz

Ar	<div></div>
X	
7	
14,14	

Total Score (points)	Answer t
	14
	14,1
	22
	9,4
	13,5
	13,6
	20
	11,6
	4,3
	25,9

5 Quiz

	7
	15,9
	24,4
	24,6
	27,3
	7,9
	14,4
	13,3
	5,3
	19,1
	9,3
	9,5
	27,3
	11,8

5 Quiz

block 7- a
6 Quiz
Correct answers
Players correct (
Question duratic
Answer Sum
Answer options
Is answer correc
Number of answ
Average time tal
Answer Deta
Players
Bianca
Brandon
Dana
Daniel
Emily
Garrett Keeney
Henry
Julian
Kent
Orion

6 Quiz

Ryan
Savannah
Shane
Will C
Willow
balin
carly
fernanda
kevy
lauren
niklas
sophie
victoria
♠William L♠

Atomic radius trend	
What's the principle of electron-electron repulsion?	
s	The prin
(%)	83,33%
on	30 secur

Summary	
	▲
st?	
ers received	
ken to answer (seconds)	

ails	
	Answer
	✗
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✗
	✓

6 Quiz

	✓
	✓
	✓
	✓
	✓
	✓
	✗
	✓
	✓
	✗
	✓
	✓
	✓
	✓
	✓
	✓

6 Quiz

principle that electron pairs orient themselves as far as possible
nds

Principle that electrons come together	◆
X	
1	
11,80	

	Score (p
Principle where protons are repelled from each other	0
The principle that electron pairs orient themselves as far as possible	770
The principle that electron pairs orient themselves as far as possible	830
The principle that electron pairs orient themselves as far as possible	1348
The principle that electron pairs orient themselves as far as possible	802
The principle that electron pairs orient themselves as far as possible	867
The principle that electron pairs orient themselves as far as possible	1293
The principle that electron pairs orient themselves as far as possible	1348
Principle where protons are repelled from each other	0
The principle that electron pairs orient themselves as far as possible	885

6 Quiz

The principle that electron pairs orient themselves as far as possible	1397
The principle that electron pairs orient themselves as far as possible	875
The principle that electron pairs orient themselves as far as possible	520
The principle that electron pairs orient themselves as far as possible	905
The principle that electron pairs orient themselves as far as possible	1287
Principle where protons are repelled from each other	0
The principle that electron pairs orient themselves as far as possible	800
The principle that electron pairs orient themselves as far as possible	812
Principle that electrons come together	0
The principle that electron pairs orient themselves as far as possible	1278
The principle that electron pairs orient themselves as far as possible	1212
The principle that electron pairs orient themselves as far as possible	1360
The principle that electron pairs orient themselves as far as possible	663
The principle that electron pairs orient themselves as far as possible	910

6 Quiz

The principle that electron pairs orient themselves as far as possible	<div><div></div></div>
✓	
20	
12,12	

oints)	Current
	3790
	3646
	5105
	6798
	2624
	3500
	5703
	6750
	2985
	4214

6 Quiz

	6895
	2770
	3433
	4244
	6091
	4410
	3793
	3848
	4415
	6598
	5260
	6875
	4127
	2860

6 Quiz

Principle where protons are repelled from each other	<input checked="" type="checkbox"/>
X	
3	
10,87	

Total Score (points)	Answer t
	9,1
	13,8
	10,2
	9,1
	11,9
	14
	12,4
	9,1
	13,9
	12,9

6 Quiz

	6,2
	7,5
	28,8
	11,7
	12,8
	9,6
	12
	11,3
	11,8
	13,3
	5,3
	8,4
	26,2
	5,4

6 Quiz

Time (seconds)	State
0	When protons and electrons combine and make a atom
0,00	X

6 Quiz

block 7- a
7 Quiz
Correct answers
Players correct (
Question duratic
Answer Sum
Answer options
Is answer correc
Number of answ
Average time tal
Answer Deta
Players
Bianca
Brandon
Dana
Daniel
Emily
Garrett Keeney
Henry
Julian
Kent
Orion

7 Quiz

Ryan
Savannah
Shane
Will C
Willow
balin
carly
fernanda
kevy
lauren
niklas
sophie
victoria
♠William L♠

Atomic radius trend	
How does electron-electron repulsion affect the size of an atom?	
s	It makes
(%)	100,00%
on	30 secur

Summary	
	▲
st?	
ers received	
ken to answer (seconds)	




Hails	
	Answer
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓

7 Quiz

	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓

7 Quiz

it larger
,
nds

It makes it larger	
	
24	
8,78	

	Score (p
It makes it larger	867
It makes it larger	920
It makes it larger	982
It makes it larger	1387
It makes it larger	828
It makes it larger	1068
It makes it larger	1302
It makes it larger	1362
It makes it larger	813
It makes it larger	1048

7 Quiz

It makes it larger	1360
It makes it larger	947
It makes it larger	992
It makes it larger	1022
It makes it larger	1363
It makes it larger	797
It makes it larger	885
It makes it larger	1048
It makes it larger	917
It makes it larger	1293
It makes it larger	1322
It makes it larger	1375
It makes it larger	1115
It makes it larger	978

7 Quiz

Stays the same	<input checked="" type="radio"/>
X	<input type="radio"/>
0	
0,00	

oints)	Current
	4657
	4566
	6087
	8185
	3452
	4568
	7005
	8112
	3798
	5262

7 Quiz

	8255
	3717
	4425
	5266
	7454
	5207
	4678
	4896
	5332
	7891
	6582
	8250
	5242
	3838

7 Quiz

It makes it smaller	<input type="checkbox"/>
X	
0	
0,00	

Total Score (points)	Answer ti
	8
	10,8
	7,1
	6,8
	16,3
	7,9
	11,9
	8,3
	11,2
	9,1

7 Quiz

	8,4
	9,2
	6,5
	10,7
	8,2
	12,2
	12,9
	3,1
	5
	12,4
	4,7
	7,5
	5,1
	7,3

7 Quiz

The chart displays the distribution of time spent on different tasks. The y-axis represents time in seconds, ranging from 0 to 100. The x-axis represents the size of the task, ranging from 0 to 100. The distribution is shown as a series of horizontal bars of varying lengths, with the longest bar at 100 seconds.

Time (seconds)	Task Size (approximate)
0	100
10	100
20	100
30	100
40	100
50	100
60	100
70	100
80	100
90	100
100	100

7 Quiz

block 7- a
8 Quiz
Correct answers
Players correct (
Question duratic
Answer Sum
Answer options
Is answer correc
Number of answ
Average time tal
Answer Deta
Players
Bianca
Brandon
Dana
Daniel
Emily
Garrett Keeney
Henry
Julian
Kent
Orion

8 Quiz

Ryan
Savannah
Shane
Will C
Willow
balin
carly
fernanda
kevy
lauren
niklas
sophie
victoria
♠William L♠

Atomic radius trend	
How does coulomb's law affect the atomic radius trend?	
s	The attra
(%)	79,17%
on	30 secur

Summary	
	▲
st?	
ers received	
ken to answer (seconds)	

ails	
	Answer
	✓
	✓
	✓
	✓
	✗
	✓
	✓
	✗
	✓
	✓

8 Quiz

	✓
	✓
	✗
	✓
	✓
	✓
	✗
	✗
	✓
	✓
	✓
	✓
	✓
	✓
	✓

8 Quiz

action gets stronger
nds

The attraction does nothing	◆
X	
1	
15,00	

	Score (p
The attraction gets stronger	823
The attraction gets stronger	1135
The attraction gets stronger	1115
The attraction gets stronger	1362
The attraction does nothing	0
The attraction gets stronger	1137
The attraction gets stronger	1293
the atomic radius doubles	0
The attraction gets stronger	973
The attraction gets stronger	978

8 Quiz

The attraction gets stronger	1400
The attraction gets stronger	898
The attraction gets weaker	0
The attraction gets stronger	973
The attraction gets stronger	1383
The attraction gets stronger	950
The attraction gets weaker	0
The attraction gets weaker	0
The attraction gets stronger	997
The attraction gets stronger	1312
The attraction gets stronger	1368
The attraction gets stronger	1395
The attraction gets stronger	1105
The attraction gets stronger	892

8 Quiz

the atomic radius doubles	<div><div></div></div>
X	
1	
14,60	

oints)	Current
	5480
	5701
	7202
	9547
	3452
	5705
	8298
	8112
	4771
	6240

8 Quiz

	9655
	4615
	4425
	6239
	8837
	6157
	4678
	4896
	6329
	9203
	7950
	9645
	6347
	4730

8 Quiz

The attraction gets stronger	<div><div></div></div>
<div><div>✓</div></div>	
19	
10,77	

Total Score (points)	Answer ti
	16,6
	3,9
	5,1
	8,3
	15
	9,8
	12,4
	14,6
	7,6
	19,3

8 Quiz

	6
	18,1
	25,8
	19,6
	7
	9
	9,7
	9,1
	6,2
	11,3
	7,9
	6,3
	11,7
	18,5

8 Quiz

Stimulus	Time (seconds)
The attraction gets weaker	14,870
The attraction gets stronger	3,000
The attraction remains the same	1,000

8 Quiz

block 7- a
9 Quiz
Correct answers
Players correct (
Question duratic
Answer Sum
Answer options
Is answer correc
Number of answ
Average time tal
Answer Deta
Players
Bianca
Brandon
Dana
Daniel
Emily
Garrett Keeney
Henry
Julian
Kent
Orion

9 Quiz

Ryan
Savannah
Shane
Will C
Willow
balin
carly
fernanda
kevy
lauren
niklas
sophie
victoria
♠William L♠

Atomic radius trend	
What happens to the trend when moving left to right across a period?	
Electronegativity	Increases
Atomic radius (%)	Decreases
Time taken to answer (seconds)	30 seconds

Summary	
Score	100%
Correct?	Yes
Answers received	1
Time taken to answer (seconds)	30

Details	
Question	Answer
What happens to the trend when moving left to right across a period?	Decreases
What happens to the trend when moving left to right across a period?	Decreases
What happens to the trend when moving left to right across a period?	Decreases
What happens to the trend when moving left to right across a period?	Decreases
What happens to the trend when moving left to right across a period?	Decreases
What happens to the trend when moving left to right across a period?	Decreases
What happens to the trend when moving left to right across a period?	Decreases
What happens to the trend when moving left to right across a period?	Decreases
What happens to the trend when moving left to right across a period?	Decreases
What happens to the trend when moving left to right across a period?	Decreases
What happens to the trend when moving left to right across a period?	Decreases

9 Quiz

	X
	X
	X
	X
	X
	X
	X
	X
	X
	X
	X
	X
	X
	X
	X
	X

iod?
egativity from left to right increases
nds

Electronegativity from left to right increases	◆
✓	
0	
0,00	

	Score (p
	0
	0
	0
Electronegativity from left to right decreases	0
	0
Electronegativity from left to right decreases	0
Electronegativity from left to right decreases	0
Electronegativity from left to right decreases	0
	0
Electronegativity from left to right decreases	0

9 Quiz

Electronegativity from left to right decreases	0
Electronegativity from left to right decreases	0
Electronegativity from left to right decreases	0
Electronegativity from left to right decreases	0
	0
Electronegativity from left to right decreases	0
Electronegativity from left to right decreases	0
Electronegativity from left to right decreases	0
Electronegativity from left to right decreases	0
Electronegativity from left to right decreases	0
Electronegativity from left to right decreases	0
Electronegativity from left to right decreases	0
	0
Electronegativity from left to right decreases	0

Electronegativity from left to right decreases	<div><div></div></div>
X	
17	
11,04	

oints)	Current
	5480
	5701
	7202
	9547
	3452
	5705
	8298
	8112
	4771
	6240

9 Quiz

	9655
	4615
	4425
	6239
	8837
	6157
	4678
	4896
	6329
	9203
	7950
	9645
	6347
	4730

9 Quiz

Electronegativity from right to left increases	<input type="checkbox"/>
X	
0	
0,00	

Total Score (points)	Answer ti
	30
	30
	30
	13,1
	30
	12,2
	14,3
	2,5
	30
	13,9

9 Quiz

	9,1
	12,6
	10,9
	15,5
	30
	16,8
	10,5
	10,6
	5,8
	14,1
	5,6
	10,1
	30
	10

[illegible]

9 Quiz

block 7- a
10 Quiz
Correct answers
Players correct (
Question duratic
Answer Sum
Answer options
Is answer correc
Number of answ
Average time tal
Answer Deta
Players
Bianca
Brandon
Dana
Daniel
Emily
Garrett Keeney
Henry
Julian
Kent
Orion

10 Quiz

Ryan
Savannah
Shane
Will C
Willow
balin
carly
fernanda
kevy
lauren
niklas
sophie
victoria
♠William L♠

Atomic radius trend	
What happens to a trend moving down a group?	
s	Electronegativity
(%)	8,33%
on	30 seconds

Summary	
	▲
st?	
ers received	
ken to answer (seconds)	

Details	
	Answer
	X
	X
	X
	X
	X
	X
	X
	✓
	X
	X

10 Quiz

	X
	X
	X
	X
	X
	X
	X
	X
	X
	X
	X
	X
	X
	X
	✓
	X

egativity of an element decreases
nds

Electronegativity of an element increases	◆
X	
13	
9,46	

	Score (p
	0
Electronegativity of an element increases	0
	0
Electronegativity of an element increases	0
	0
Electronegativity of an element increases	0
	0
Electronegativity of an element decreases	895
Electronegativity of an element increases	0
	0

10 Quiz

Electronegativity of an element increases	0
Electronegativity of an element increases	0
Electronegativity of an element increases	0
	0
	0
Electronegativity of an element increases	0
Electronegativity of an element increases	0
Electronegativity of an element increases	0
Electronegativity of an element increases	0
	0
Electronegativity of an element increases	0
Electronegativity of an element increases	0
Electronegativity of an element decreases	903
	0

Electronegativity of an compound decreases	<div><div></div></div>
X	
0	
0,00	

oints)	Current
	5480
	5701
	7202
	9547
	3452
	5705
	8298
	9007
	4771
	6240

10 Quiz

	9655
	4615
	4425
	6239
	8837
	6157
	4678
	4896
	6329
	9203
	7950
	9645
	7250
	4730

10 Quiz

Electronegativity of an compound increases	<input type="checkbox"/>
X	
0	
0,00	

Total Score (points)	Answer ti
	30
	14,1
	30
	10,9
	30
	11
	30
	6,3
	6,9
	30

10 Quiz

	7,2
	13,3
	12,1
	30
	30
	4,1
	13,4
	12,6
	4,5
	30
	5,4
	7,5
	5,8
	30

[illegible]

10 Quiz

Question Number
1 Quiz
1 Quiz
1 Quiz
1 Quiz
1 Quiz
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10 Quiz

Question
what is the radius?
what is the radius?
what is the radius?
what is the radius?
what is the radius?
what is the radius?
what is the radius?
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what is the radius?
what is the radius?
what is the radius?
what is the radius?
what is the radius?
what is the radius?
what is the radius?
what is the radius?
what is the radius?
How is the atomic radius measures?
How is the atomic radius measures?
How is the atomic radius measures?
How is the atomic radius measures?
How is the atomic radius measures?
How is the atomic radius measures?
How is the atomic radius measures?

How is the atomic radius measures?
How is the atomic radius measures?
How is the atomic radius measures?
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How is the atomic radius measures?
How is the atomic radius measures?
How is the atomic radius measures?
How is the atomic radius measures?
How is the atomic radius measures?
How is the atomic radius measures?

How is the atomic radius measures?
The nuclear force is the force between nucleons in the nucleus that holds the nucleus together
The nuclear force is the force between nucleons in the nucleus that holds the nucleus together
The nuclear force is the force between nucleons in the nucleus that holds the nucleus together
The nuclear force is the force between nucleons in the nucleus that holds the nucleus together
The nuclear force is the force between nucleons in the nucleus that holds the nucleus together
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The nuclear force is the force between nucleons in the nucleus that holds the nucleus together

The nuclear force is the force between nucleons in the nucleus that holds the nucleus together
The nuclear force is the force between nucleons in the nucleus that holds the nucleus together
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The nuclear force is the force between nucleons in the nucleus that holds the nucleus together
The nuclear force is the force between nucleons in the nucleus that holds the nucleus together
The nuclear force is the force between nucleons in the nucleus that holds the nucleus together
The nuclear force is the force between nucleons in the nucleus that holds the nucleus together
The nuclear force holds protons together because opposite charges attract
The nuclear force holds protons together because opposite charges attract
The nuclear force holds protons together because opposite charges attract
The nuclear force holds protons together because opposite charges attract
The nuclear force holds protons together because opposite charges attract
The nuclear force holds protons together because opposite charges attract
The nuclear force holds protons together because opposite charges attract

The nuclear force holds protons together because opposite charges attract
The nuclear force holds protons together because opposite charges attract
The nuclear force holds protons together because opposite charges attract
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The nuclear force holds protons together because opposite charges attract
The nuclear force holds protons together because opposite charges attract
The nuclear force holds protons together because opposite charges attract
The nuclear force holds protons together because opposite charges attract

The nuclear force holds protons together because opposite charges attract
Through nuclear force, which atom is the largest size?
Through nuclear force, which atom is the largest size?
Through nuclear force, which atom is the largest size?
Through nuclear force, which atom is the largest size?
Through nuclear force, which atom is the largest size?
Through nuclear force, which atom is the largest size?
Through nuclear force, which atom is the largest size?
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Through nuclear force, which atom is the largest size?
Through nuclear force, which atom is the largest size?

Through nuclear force, which atom is the largest size?
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Through nuclear force, which atom is the largest size?
Through nuclear force, which atom is the largest size?
Through nuclear force, which atom is the largest size?
Through nuclear force, which atom is the largest size?
Through nuclear force, which atom is the largest size?
Through nuclear force, which atom is the largest size?
What's the principle of electron-electron repulsion?
What's the principle of electron-electron repulsion?
What's the principle of electron-electron repulsion?
What's the principle of electron-electron repulsion?
What's the principle of electron-electron repulsion?
What's the principle of electron-electron repulsion?
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What's the principle of electron-electron repulsion?
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What's the principle of electron-electron repulsion?
What's the principle of electron-electron repulsion?
What's the principle of electron-electron repulsion?
What's the principle of electron-electron repulsion?

What's the principle of electron-electron repulsion?
How does electron-electron repulsion affect the size of an atom?
How does electron-electron repulsion affect the size of an atom?
How does electron-electron repulsion affect the size of an atom?
How does electron-electron repulsion affect the size of an atom?
How does electron-electron repulsion affect the size of an atom?
How does electron-electron repulsion affect the size of an atom?
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How does electron-electron repulsion affect the size of an atom?
How does electron-electron repulsion affect the size of an atom?
How does electron-electron repulsion affect the size of an atom?
How does electron-electron repulsion affect the size of an atom?
How does electron-electron repulsion affect the size of an atom?
How does coulomb's law affect the atomic radius trend?
How does coulomb's law affect the atomic radius trend?
How does coulomb's law affect the atomic radius trend?
How does coulomb's law affect the atomic radius trend?
How does coulomb's law affect the atomic radius trend?
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How does coulomb's law affect the atomic radius trend?
How does coulomb's law affect the atomic radius trend?
How does coulomb's law affect the atomic radius trend?

How does coulomb's law affect the atomic radius trend?
What happens to the trend when moving left to right across a period?
What happens to the trend when moving left to right across a period?
What happens to the trend when moving left to right across a period?
What happens to the trend when moving left to right across a period?
What happens to the trend when moving left to right across a period?
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What happens to the trend when moving left to right across a period?
What happens to the trend when moving left to right across a period?
What happens to the trend when moving left to right across a period?
What happens to the trend when moving left to right across a period?
What happens to the trend when moving left to right across a period?
What happens to a trend moving down a group?
What happens to a trend moving down a group?
What happens to a trend moving down a group?
What happens to a trend moving down a group?
What happens to a trend moving down a group?
What happens to a trend moving down a group?
What happens to a trend moving down a group?

What happens to a trend moving down a group?
What happens to a trend moving down a group?
What happens to a trend moving down a group?
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What happens to a trend moving down a group?
What happens to a trend moving down a group?
What happens to a trend moving down a group?
What happens to a trend moving down a group?
What happens to a trend moving down a group?

What happens to a trend moving down a group?

Answer 1	Answer 2
the distance around the circle	half the diameter of a circle
the distance around the circle	half the diameter of a circle
the distance around the circle	half the diameter of a circle
the distance around the circle	half the diameter of a circle
the distance around the circle	half the diameter of a circle
the distance around the circle	half the diameter of a circle
the distance around the circle	half the diameter of a circle
the distance around the circle	half the diameter of a circle
the distance around the circle	half the diameter of a circle
the distance around the circle	half the diameter of a circle
the distance around the circle	half the diameter of a circle
the distance around the circle	half the diameter of a circle
the distance around the circle	half the diameter of a circle
the distance around the circle	half the diameter of a circle
the distance around the circle	half the diameter of a circle

the distance around the circle	half the diameter of a circle
the distance around the circle	half the diameter of a circle
the distance around the circle	half the diameter of a circle
the distance around the circle	half the diameter of a circle
the distance around the circle	half the diameter of a circle
the distance around the circle	half the diameter of a circle
the distance around the circle	half the diameter of a circle
the distance around the circle	half the diameter of a circle
the distance around the circle	half the diameter of a circle
the distance from the center of a nucleus to the outside of the atom	the distance from the outside of one atom to the outside of another
the distance from the center of a nucleus to the outside of the atom	the distance from the outside of one atom to the outside of another
the distance from the center of a nucleus to the outside of the atom	the distance from the outside of one atom to the outside of another
the distance from the center of a nucleus to the outside of the atom	the distance from the outside of one atom to the outside of another
the distance from the center of a nucleus to the outside of the atom	the distance from the outside of one atom to the outside of another
the distance from the center of a nucleus to the outside of the atom	the distance from the outside of one atom to the outside of another
the distance from the center of a nucleus to the outside of the atom	the distance from the outside of one atom to the outside of another

[illegible]

the distance from the center of a nucleus to the outside of the atom	the distance from the outside of one atom to the outside of another
False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True

RawReportData Data

False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True

RawReportData Data

False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True
False	True

RawReportData Data

False	True
P	Mg
P	Mg
P	Mg
P	Mg
P	Mg
P	Mg
P	Mg
P	Mg
P	Mg
P	Mg
P	Mg
P	Mg
P	Mg
P	Mg
P	Mg

P	Mg
P	Mg
P	Mg
P	Mg
P	Mg
P	Mg
P	Mg
P	Mg
P	Mg
Principle that electrons come together	The principle that electron pairs orient themselves as far as possible
Principle that electrons come together	The principle that electron pairs orient themselves as far as possible
Principle that electrons come together	The principle that electron pairs orient themselves as far as possible
Principle that electrons come together	The principle that electron pairs orient themselves as far as possible
Principle that electrons come together	The principle that electron pairs orient themselves as far as possible
Principle that electrons come together	The principle that electron pairs orient themselves as far as possible
Principle that electrons come together	The principle that electron pairs orient themselves as far as possible

Principle that electrons come together	The principle that electron pairs orient themselves as far as possible
Principle that electrons come together	The principle that electron pairs orient themselves as far as possible
Principle that electrons come together	The principle that electron pairs orient themselves as far as possible
Principle that electrons come together	The principle that electron pairs orient themselves as far as possible
Principle that electrons come together	The principle that electron pairs orient themselves as far as possible
Principle that electrons come together	The principle that electron pairs orient themselves as far as possible
Principle that electrons come together	The principle that electron pairs orient themselves as far as possible
Principle that electrons come together	The principle that electron pairs orient themselves as far as possible
Principle that electrons come together	The principle that electron pairs orient themselves as far as possible
Principle that electrons come together	The principle that electron pairs orient themselves as far as possible
Principle that electrons come together	The principle that electron pairs orient themselves as far as possible
Principle that electrons come together	The principle that electron pairs orient themselves as far as possible
Principle that electrons come together	The principle that electron pairs orient themselves as far as possible
Principle that electrons come together	The principle that electron pairs orient themselves as far as possible
Principle that electrons come together	The principle that electron pairs orient themselves as far as possible
Principle that electrons come together	The principle that electron pairs orient themselves as far as possible

Principle that electrons come together	The principle that electron pairs orient themselves as far as possible
It makes it larger	Stays the same
It makes it larger	Stays the same
It makes it larger	Stays the same
It makes it larger	Stays the same
It makes it larger	Stays the same
It makes it larger	Stays the same
It makes it larger	Stays the same
It makes it larger	Stays the same
It makes it larger	Stays the same
It makes it larger	Stays the same
It makes it larger	Stays the same
It makes it larger	Stays the same
It makes it larger	Stays the same
It makes it larger	Stays the same
It makes it larger	Stays the same

It makes it larger	Stays the same
It makes it larger	Stays the same
It makes it larger	Stays the same
It makes it larger	Stays the same
It makes it larger	Stays the same
It makes it larger	Stays the same
It makes it larger	Stays the same
It makes it larger	Stays the same
It makes it larger	Stays the same
The attraction does nothing	the atomic radius doubles
The attraction does nothing	the atomic radius doubles
The attraction does nothing	the atomic radius doubles
The attraction does nothing	the atomic radius doubles
The attraction does nothing	the atomic radius doubles
The attraction does nothing	the atomic radius doubles
The attraction does nothing	the atomic radius doubles

The attraction does nothing	the atomic radius doubles
The attraction does nothing	the atomic radius doubles
The attraction does nothing	the atomic radius doubles
The attraction does nothing	the atomic radius doubles
The attraction does nothing	the atomic radius doubles
The attraction does nothing	the atomic radius doubles
The attraction does nothing	the atomic radius doubles
The attraction does nothing	the atomic radius doubles
The attraction does nothing	the atomic radius doubles
The attraction does nothing	the atomic radius doubles
The attraction does nothing	the atomic radius doubles
The attraction does nothing	the atomic radius doubles
The attraction does nothing	the atomic radius doubles
The attraction does nothing	the atomic radius doubles
The attraction does nothing	the atomic radius doubles
The attraction does nothing	the atomic radius doubles

The attraction does nothing	the atomic radius doubles
Electronegativity from left to right increases	Electronegativity from left to right decreases
Electronegativity from left to right increases	Electronegativity from left to right decreases
Electronegativity from left to right increases	Electronegativity from left to right decreases
Electronegativity from left to right increases	Electronegativity from left to right decreases
Electronegativity from left to right increases	Electronegativity from left to right decreases
Electronegativity from left to right increases	Electronegativity from left to right decreases
Electronegativity from left to right increases	Electronegativity from left to right decreases
Electronegativity from left to right increases	Electronegativity from left to right decreases
Electronegativity from left to right increases	Electronegativity from left to right decreases
Electronegativity from left to right increases	Electronegativity from left to right decreases
Electronegativity from left to right increases	Electronegativity from left to right decreases
Electronegativity from left to right increases	Electronegativity from left to right decreases
Electronegativity from left to right increases	Electronegativity from left to right decreases
Electronegativity from left to right increases	Electronegativity from left to right decreases
Electronegativity from left to right increases	Electronegativity from left to right decreases

Electronegativity from left to right increases	Electronegativity from left to right decreases
Electronegativity from left to right increases	Electronegativity from left to right decreases
Electronegativity from left to right increases	Electronegativity from left to right decreases
Electronegativity from left to right increases	Electronegativity from left to right decreases
Electronegativity from left to right increases	Electronegativity from left to right decreases
Electronegativity from left to right increases	Electronegativity from left to right decreases
Electronegativity from left to right increases	Electronegativity from left to right decreases
Electronegativity from left to right increases	Electronegativity from left to right decreases
Electronegativity from left to right increases	Electronegativity from left to right decreases
Electronegativity of an element increases	Electronegativity of an compound decreases
Electronegativity of an element increases	Electronegativity of an compound decreases
Electronegativity of an element increases	Electronegativity of an compound decreases
Electronegativity of an element increases	Electronegativity of an compound decreases
Electronegativity of an element increases	Electronegativity of an compound decreases
Electronegativity of an element increases	Electronegativity of an compound decreases
Electronegativity of an element increases	Electronegativity of an compound decreases

Electronegativity of an element increases	Electronegativity of an compound decreases
Electronegativity of an element increases	Electronegativity of an compound decreases
Electronegativity of an element increases	Electronegativity of an compound decreases
Electronegativity of an element increases	Electronegativity of an compound decreases
Electronegativity of an element increases	Electronegativity of an compound decreases
Electronegativity of an element increases	Electronegativity of an compound decreases
Electronegativity of an element increases	Electronegativity of an compound decreases
Electronegativity of an element increases	Electronegativity of an compound decreases
Electronegativity of an element increases	Electronegativity of an compound decreases
Electronegativity of an element increases	Electronegativity of an compound decreases
Electronegativity of an element increases	Electronegativity of an compound decreases
Electronegativity of an element increases	Electronegativity of an compound decreases
Electronegativity of an element increases	Electronegativity of an compound decreases
Electronegativity of an element increases	Electronegativity of an compound decreases
Electronegativity of an element increases	Electronegativity of an compound decreases
Electronegativity of an element increases	Electronegativity of an compound decreases

Electronegativity of an element increases	Electronegativity of an compound decreases
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Answer 3	Answer 4
the distance from one circle to another	the distance between robin and his dead parents
the distance from one circle to another	the distance between robin and his dead parents
the distance from one circle to another	the distance between robin and his dead parents
the distance from one circle to another	the distance between robin and his dead parents
the distance from one circle to another	the distance between robin and his dead parents
the distance from one circle to another	the distance between robin and his dead parents
the distance from one circle to another	the distance between robin and his dead parents
the distance from one circle to another	the distance between robin and his dead parents
the distance from one circle to another	the distance between robin and his dead parents
the distance from one circle to another	the distance between robin and his dead parents
the distance from one circle to another	the distance between robin and his dead parents
the distance from one circle to another	the distance between robin and his dead parents
the distance from one circle to another	the distance between robin and his dead parents
the distance from one circle to another	the distance between robin and his dead parents
the distance from one circle to another	the distance between robin and his dead parents

the distance from one circle to another	the distance between robin and his dead parents
the distance from one circle to another	the distance between robin and his dead parents
the distance from one circle to another	the distance between robin and his dead parents
the distance from one circle to another	the distance between robin and his dead parents
the distance from one circle to another	the distance between robin and his dead parents
the distance from one circle to another	the distance between robin and his dead parents
the distance from one circle to another	the distance between robin and his dead parents
the distance from one circle to another	the distance between robin and his dead parents
the distance from one circle to another	the distance between robin and his dead parents
Half of the distance between the nuclei of two atoms	the total amount around two different atoms
Half of the distance between the nuclei of two atoms	the total amount around two different atoms
Half of the distance between the nuclei of two atoms	the total amount around two different atoms
Half of the distance between the nuclei of two atoms	the total amount around two different atoms
Half of the distance between the nuclei of two atoms	the total amount around two different atoms
Half of the distance between the nuclei of two atoms	the total amount around two different atoms
Half of the distance between the nuclei of two atoms	the total amount around two different atoms

Half of the distance between the nuclei of two atoms	the total amount around two different atoms
Half of the distance between the nuclei of two atoms	the total amount around two different atoms
Half of the distance between the nuclei of two atoms	the total amount around two different atoms
Half of the distance between the nuclei of two atoms	the total amount around two different atoms
Half of the distance between the nuclei of two atoms	the total amount around two different atoms
Half of the distance between the nuclei of two atoms	the total amount around two different atoms
Half of the distance between the nuclei of two atoms	the total amount around two different atoms
Half of the distance between the nuclei of two atoms	the total amount around two different atoms
Half of the distance between the nuclei of two atoms	the total amount around two different atoms
Half of the distance between the nuclei of two atoms	the total amount around two different atoms
Half of the distance between the nuclei of two atoms	the total amount around two different atoms
Half of the distance between the nuclei of two atoms	the total amount around two different atoms
Half of the distance between the nuclei of two atoms	the total amount around two different atoms
Half of the distance between the nuclei of two atoms	the total amount around two different atoms
Half of the distance between the nuclei of two atoms	the total amount around two different atoms

Half of the distance between the nuclei of two atoms	the total amount around two different atoms

RawReportData Data

Ar	Na
Ar	Na
Ar	Na
Ar	Na
Ar	Na
Ar	Na
Ar	Na
Ar	Na
Ar	Na
Ar	Na
Ar	Na
Ar	Na
Ar	Na
Ar	Na
Ar	Na

Ar	Na
Ar	Na
Ar	Na
Ar	Na
Ar	Na
Ar	Na
Ar	Na
Ar	Na
Ar	Na
Principle where protons are repelled from each other	When protons and electrons combine and make a atom
Principle where protons are repelled from each other	When protons and electrons combine and make a atom
Principle where protons are repelled from each other	When protons and electrons combine and make a atom
Principle where protons are repelled from each other	When protons and electrons combine and make a atom
Principle where protons are repelled from each other	When protons and electrons combine and make a atom
Principle where protons are repelled from each other	When protons and electrons combine and make a atom
Principle where protons are repelled from each other	When protons and electrons combine and make a atom

Principle where protons are repelled from each other	When protons and electrons combine and make a atom
Principle where protons are repelled from each other	When protons and electrons combine and make a atom
Principle where protons are repelled from each other	When protons and electrons combine and make a atom
Principle where protons are repelled from each other	When protons and electrons combine and make a atom
Principle where protons are repelled from each other	When protons and electrons combine and make a atom
Principle where protons are repelled from each other	When protons and electrons combine and make a atom
Principle where protons are repelled from each other	When protons and electrons combine and make a atom
Principle where protons are repelled from each other	When protons and electrons combine and make a atom
Principle where protons are repelled from each other	When protons and electrons combine and make a atom
Principle where protons are repelled from each other	When protons and electrons combine and make a atom
Principle where protons are repelled from each other	When protons and electrons combine and make a atom
Principle where protons are repelled from each other	When protons and electrons combine and make a atom
Principle where protons are repelled from each other	When protons and electrons combine and make a atom
Principle where protons are repelled from each other	When protons and electrons combine and make a atom
Principle where protons are repelled from each other	When protons and electrons combine and make a atom

Principle where protons are repelled from each other	When protons and electrons combine and make a atom
It makes it smaller	the size evens out
It makes it smaller	the size evens out
It makes it smaller	the size evens out
It makes it smaller	the size evens out
It makes it smaller	the size evens out
It makes it smaller	the size evens out
It makes it smaller	the size evens out
It makes it smaller	the size evens out
It makes it smaller	the size evens out
It makes it smaller	the size evens out
It makes it smaller	the size evens out
It makes it smaller	the size evens out
It makes it smaller	the size evens out
It makes it smaller	the size evens out
It makes it smaller	the size evens out

It makes it smaller	the size evens out
It makes it smaller	the size evens out
It makes it smaller	the size evens out
It makes it smaller	the size evens out
It makes it smaller	the size evens out
It makes it smaller	the size evens out
It makes it smaller	the size evens out
It makes it smaller	the size evens out
It makes it smaller	the size evens out
The attraction gets stronger	The attraction gets weaker
The attraction gets stronger	The attraction gets weaker
The attraction gets stronger	The attraction gets weaker
The attraction gets stronger	The attraction gets weaker
The attraction gets stronger	The attraction gets weaker
The attraction gets stronger	The attraction gets weaker
The attraction gets stronger	The attraction gets weaker

The attraction gets stronger	The attraction gets weaker
The attraction gets stronger	The attraction gets weaker
The attraction gets stronger	The attraction gets weaker
The attraction gets stronger	The attraction gets weaker
The attraction gets stronger	The attraction gets weaker
The attraction gets stronger	The attraction gets weaker
The attraction gets stronger	The attraction gets weaker
The attraction gets stronger	The attraction gets weaker
The attraction gets stronger	The attraction gets weaker
The attraction gets stronger	The attraction gets weaker
The attraction gets stronger	The attraction gets weaker
The attraction gets stronger	The attraction gets weaker
The attraction gets stronger	The attraction gets weaker
The attraction gets stronger	The attraction gets weaker
The attraction gets stronger	The attraction gets weaker
The attraction gets stronger	The attraction gets weaker

The attraction gets stronger	The attraction gets weaker
Electronegativity from right to left increases	Electronegativity from right to left decreases
Electronegativity from right to left increases	Electronegativity from right to left decreases
Electronegativity from right to left increases	Electronegativity from right to left decreases
Electronegativity from right to left increases	Electronegativity from right to left decreases
Electronegativity from right to left increases	Electronegativity from right to left decreases
Electronegativity from right to left increases	Electronegativity from right to left decreases
Electronegativity from right to left increases	Electronegativity from right to left decreases
Electronegativity from right to left increases	Electronegativity from right to left decreases
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Electronegativity from right to left increases	Electronegativity from right to left decreases
Electronegativity from right to left increases	Electronegativity from right to left decreases
Electronegativity from right to left increases	Electronegativity from right to left decreases
Electronegativity from right to left increases	Electronegativity from right to left decreases
Electronegativity from right to left increases	Electronegativity from right to left decreases

Electronegativity from right to left increases	Electronegativity from right to left decreases
Electronegativity from right to left increases	Electronegativity from right to left decreases
Electronegativity from right to left increases	Electronegativity from right to left decreases
Electronegativity from right to left increases	Electronegativity from right to left decreases
Electronegativity from right to left increases	Electronegativity from right to left decreases
Electronegativity from right to left increases	Electronegativity from right to left decreases
Electronegativity from right to left increases	Electronegativity from right to left decreases
Electronegativity from right to left increases	Electronegativity from right to left decreases
Electronegativity from right to left increases	Electronegativity from right to left decreases
Electronegativity of an compound increases	Electronegativity of an element decreases
Electronegativity of an compound increases	Electronegativity of an element decreases
Electronegativity of an compound increases	Electronegativity of an element decreases
Electronegativity of an compound increases	Electronegativity of an element decreases
Electronegativity of an compound increases	Electronegativity of an element decreases
Electronegativity of an compound increases	Electronegativity of an element decreases
Electronegativity of an compound increases	Electronegativity of an element decreases

Electronegativity of an compound increases	Electronegativity of an element decreases
Electronegativity of an compound increases	Electronegativity of an element decreases
Electronegativity of an compound increases	Electronegativity of an element decreases
Electronegativity of an compound increases	Electronegativity of an element decreases
Electronegativity of an compound increases	Electronegativity of an element decreases
Electronegativity of an compound increases	Electronegativity of an element decreases
Electronegativity of an compound increases	Electronegativity of an element decreases
Electronegativity of an compound increases	Electronegativity of an element decreases
Electronegativity of an compound increases	Electronegativity of an element decreases
Electronegativity of an compound increases	Electronegativity of an element decreases
Electronegativity of an compound increases	Electronegativity of an element decreases
Electronegativity of an compound increases	Electronegativity of an element decreases
Electronegativity of an compound increases	Electronegativity of an element decreases
Electronegativity of an compound increases	Electronegativity of an element decreases
Electronegativity of an compound increases	Electronegativity of an element decreases
Electronegativity of an compound increases	Electronegativity of an element decreases

Electronegativity of an compound increases	Electronegativity of an element decreases
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Correct Answers	Time Allotted to Answer (seconds)
half the diameter of a circle	30
half the diameter of a circle	30
half the diameter of a circle	30
half the diameter of a circle	30
half the diameter of a circle	30
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half the diameter of a circle	30
Half of the distance between the nuclei of two atoms	30
Half of the distance between the nuclei of two atoms	30
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Half of the distance between the nuclei of two atoms	30
Half of the distance between the nuclei of two atoms	30

Half of the distance between the nuclei of two atoms	30
True	30
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RawReportData Data

True	30
True	30
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False	30
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False	30

RawReportData Data

False	30
False	30
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RawReportData Data

False	30
Na	30
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The principle that electron pairs orient themselves as far as possible	30
The principle that electron pairs orient themselves as far as possible	30
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The principle that electron pairs orient themselves as far as possible	30
It makes it larger	30
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The attraction gets stronger	30
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The attraction gets stronger	30
Electronegativity from left to right increases	30
Electronegativity from left to right increases	30
Electronegativity from left to right increases	30
Electronegativity from left to right increases	30
Electronegativity from left to right increases	30
Electronegativity from left to right increases	30
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Electronegativity from left to right increases	30
Electronegativity of an element decreases	30
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Electronegativity of an element decreases	30
Electronegativity of an element decreases	30

Electronegativity of an element decreases	30
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Players
Bianca
Brandon
Dana
Daniel
Emily
Garrett Keeney
Henry
Julian
Kent
Orion
Ryan
Savannah
Shane
Will C
Willow

balin
carly
fernanda
kevy
lauren
niklas
sophie
victoria
♠William L♠
Bianca
Brandon
Dana
Daniel
Emily
Garrett Keeney
Henry

Julian
Kent
Orion
Ryan
Savannah
Shane
Will C
Willow
balin
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kevy
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niklas
sophie
victoria

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Will C
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victoria
♠William L♠
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Dana
Daniel
Emily
Garrett Keeney
Henry

Julian
Kent
Orion
Ryan
Savannah
Shane
Will C
Willow
balin
carly
fernanda
kevy
lauren
niklas
sophie
victoria

♠William L♠

Answer	Correct / Incorrect	Correct
half the diameter of a circle	Correct	1
half the diameter of a circle	Correct	1
half the diameter of a circle	Correct	1
half the diameter of a circle	Correct	1
half the diameter of a circle	Correct	1
half the diameter of a circle	Correct	1
half the diameter of a circle	Correct	1
half the diameter of a circle	Correct	1
half the diameter of a circle	Correct	1
half the diameter of a circle	Correct	1
half the diameter of a circle	Correct	1
half the diameter of a circle	Correct	1
half the diameter of a circle	Correct	1
half the diameter of a circle	Correct	1
half the diameter of a circle	Correct	1

half the diameter of a circle	Correct	1
half the diameter of a circle	Correct	1
half the diameter of a circle	Correct	1
half the diameter of a circle	Correct	1
half the diameter of a circle	Correct	1
half the diameter of a circle	Correct	1
half the diameter of a circle	Correct	1
half the diameter of a circle	Correct	1
half the diameter of a circle	Correct	1
Half of the distance between the nuclei of two atoms	Correct	1
Half of the distance between the nuclei of two atoms	Correct	1
Half of the distance between the nuclei of two atoms	Correct	1
Half of the distance between the nuclei of two atoms	Correct	1
the distance from the center of a nucleus to the outside of the atom	Incorrect	0
the distance from the outside of one atom to the outside of another	Incorrect	0
Half of the distance between the nuclei of two atoms	Correct	1

Half of the distance between the nuclei of two atoms	Correct	1
the distance from the center of a nucleus to the outside of the atom	Incorrect	0
Half of the distance between the nuclei of two atoms	Correct	1
Half of the distance between the nuclei of two atoms	Correct	1
the distance from the outside of one atom to the outside of another	Incorrect	0
Half of the distance between the nuclei of two atoms	Correct	1
Half of the distance between the nuclei of two atoms	Correct	1
Half of the distance between the nuclei of two atoms	Correct	1
Half of the distance between the nuclei of two atoms	Correct	1
Half of the distance between the nuclei of two atoms	Correct	1
Half of the distance between the nuclei of two atoms	Correct	1
Half of the distance between the nuclei of two atoms	Correct	1
Half of the distance between the nuclei of two atoms	Correct	1
the distance from the center of a nucleus to the outside of the atom	Incorrect	0
Half of the distance between the nuclei of two atoms	Correct	1
Half of the distance between the nuclei of two atoms	Correct	1

the distance from the outside of one atom to the outside of another	Incorrect	0
True	Correct	1
True	Correct	1
True	Correct	1
True	Correct	1
True	Correct	1
True	Correct	1
True	Correct	1
True	Correct	1
True	Correct	1
True	Correct	1
True	Correct	1
True	Correct	1
True	Correct	1
True	Correct	1
True	Correct	1

RawReportData Data

True	Correct	1
True	Correct	1
True	Correct	1
True	Correct	1
True	Correct	1
True	Correct	1
True	Correct	1
True	Correct	1
True	Correct	1
True	Correct	1
False	Correct	1
True	Incorrect	0
False	Correct	1
False	Correct	1
True	Incorrect	0
True	Incorrect	0
False	Correct	1

RawReportData Data

False	Correct	1
False	Correct	1
True	Incorrect	0
False	Correct	1
True	Incorrect	0
True	Incorrect	0
True	Incorrect	0
False	Correct	1
False	Correct	1
True	Incorrect	0
True	Incorrect	0
False	Correct	1
False	Correct	1
False	Correct	1
False	Correct	1
True	Incorrect	0

RawReportData Data

True	Incorrect	0
Ar	Incorrect	0
Ar	Incorrect	0
Mg	Incorrect	0
Na	Correct	1
Ar	Incorrect	0
Na	Correct	1
Na	Correct	1
Na	Correct	1
P	Incorrect	0
Na	Correct	1
Na	Correct	1
Ar	Incorrect	0
Ar	Incorrect	0
Na	Correct	1
Na	Correct	1

RawReportData Data

Mg	Incorrect	0
Mg	Incorrect	0
Mg	Incorrect	0
Ar	Incorrect	0
Na	Correct	1
Na	Correct	1
Na	Correct	1
Na	Correct	1
Ar	Incorrect	0
Principle where protons are repelled from each other	Incorrect	0
The principle that electron pairs orient themselves as far as possible	Correct	1
The principle that electron pairs orient themselves as far as possible	Correct	1
The principle that electron pairs orient themselves as far as possible	Correct	1
The principle that electron pairs orient themselves as far as possible	Correct	1
The principle that electron pairs orient themselves as far as possible	Correct	1
The principle that electron pairs orient themselves as far as possible	Correct	1

The principle that electron pairs orient themselves as far as possible	Correct	1
Principle where protons are repelled from each other	Incorrect	0
The principle that electron pairs orient themselves as far as possible	Correct	1
The principle that electron pairs orient themselves as far as possible	Correct	1
The principle that electron pairs orient themselves as far as possible	Correct	1
The principle that electron pairs orient themselves as far as possible	Correct	1
The principle that electron pairs orient themselves as far as possible	Correct	1
The principle that electron pairs orient themselves as far as possible	Correct	1
Principle where protons are repelled from each other	Incorrect	0
The principle that electron pairs orient themselves as far as possible	Correct	1
The principle that electron pairs orient themselves as far as possible	Correct	1
Principle that electrons come together	Incorrect	0
The principle that electron pairs orient themselves as far as possible	Correct	1
The principle that electron pairs orient themselves as far as possible	Correct	1
The principle that electron pairs orient themselves as far as possible	Correct	1
The principle that electron pairs orient themselves as far as possible	Correct	1

RawReportData Data

The principle that electron pairs orient themselves as far as possible	Correct	1
It makes it larger	Correct	1
It makes it larger	Correct	1
It makes it larger	Correct	1
It makes it larger	Correct	1
It makes it larger	Correct	1
It makes it larger	Correct	1
It makes it larger	Correct	1
It makes it larger	Correct	1
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It makes it larger	Correct	1
It makes it larger	Correct	1
It makes it larger	Correct	1
It makes it larger	Correct	1
It makes it larger	Correct	1

RawReportData Data

It makes it larger	Correct	1
It makes it larger	Correct	1
It makes it larger	Correct	1
It makes it larger	Correct	1
It makes it larger	Correct	1
It makes it larger	Correct	1
It makes it larger	Correct	1
It makes it larger	Correct	1
It makes it larger	Correct	1
It makes it larger	Correct	1
The attraction gets stronger	Correct	1
The attraction gets stronger	Correct	1
The attraction gets stronger	Correct	1
The attraction gets stronger	Correct	1
The attraction does nothing	Incorrect	0
The attraction gets stronger	Correct	1
The attraction gets stronger	Correct	1

RawReportData Data

the atomic radius doubles	Incorrect	0
The attraction gets stronger	Correct	1
The attraction gets stronger	Correct	1
The attraction gets stronger	Correct	1
The attraction gets stronger	Correct	1
The attraction gets weaker	Incorrect	0
The attraction gets stronger	Correct	1
The attraction gets stronger	Correct	1
The attraction gets stronger	Correct	1
The attraction gets weaker	Incorrect	0
The attraction gets weaker	Incorrect	0
The attraction gets stronger	Correct	1
The attraction gets stronger	Correct	1
The attraction gets stronger	Correct	1
The attraction gets stronger	Correct	1
The attraction gets stronger	Correct	1

The attraction gets stronger	Correct	1
	Incorrect	0
	Incorrect	0
	Incorrect	0
Electronegativity from left to right decreases	Incorrect	0
	Incorrect	0
Electronegativity from left to right decreases	Incorrect	0
Electronegativity from left to right decreases	Incorrect	0
Electronegativity from left to right decreases	Incorrect	0
	Incorrect	0
Electronegativity from left to right decreases	Incorrect	0
Electronegativity from left to right decreases	Incorrect	0
Electronegativity from left to right decreases	Incorrect	0
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Electronegativity from left to right decreases	Incorrect	0
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Electronegativity from left to right decreases	Incorrect	0
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Electronegativity from left to right decreases	Incorrect	0
	Incorrect	0
Electronegativity from left to right decreases	Incorrect	0
	Incorrect	0
Electronegativity of an element increases	Incorrect	0
	Incorrect	0
Electronegativity of an element increases	Incorrect	0
	Incorrect	0
Electronegativity of an element increases	Incorrect	0
	Incorrect	0

Electronegativity of an element decreases	Correct	1
Electronegativity of an element increases	Incorrect	0
	Incorrect	0
Electronegativity of an element increases	Incorrect	0
Electronegativity of an element increases	Incorrect	0
Electronegativity of an element increases	Incorrect	0
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Electronegativity of an element increases	Incorrect	0
Electronegativity of an element increases	Incorrect	0
Electronegativity of an element increases	Incorrect	0
Electronegativity of an element increases	Incorrect	0
	Incorrect	0
Electronegativity of an element increases	Incorrect	0
Electronegativity of an element increases	Incorrect	0
Electronegativity of an element decreases	Correct	1

RawReportData Data

	Incorrect	0
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RawReportData Data

Incorrect	Score (points)	Score without Answer Streak Bonus (points)
0	838	838
0	860	860
0	953	953
0	897	897
0	960	960
0	905	905
0	663	663
0	955	955
0	940	940
0	940	940
0	928	928
0	922	922
0	803	803
0	865	865
0	765	765

RawReportData Data

0	932	932
0	942	942
0	957	957
0	933	933
0	843	843
0	958	958
0	962	962
0	873	873
0	970	970
0	732	632
0	883	783
0	897	797
0	945	845
1	0	0
1	0	0
0	735	635

RawReportData Data

0	893	793
1	0	0
0	858	758
0	925	825
1	0	0
0	935	835
0	917	817
0	947	847
0	1018	918
0	988	888
0	1007	907
0	1027	927
0	990	890
1	0	0
0	890	790
0	908	808

RawReportData Data

1	0	0
0	1085	885
0	1133	933
0	1177	977
0	1147	947
0	862	862
0	955	955
0	1050	850
0	1105	905
0	972	972
0	963	763
0	1177	977
0	973	973
0	1175	975
0	967	767
0	940	740

RawReportData Data

0	1182	982
0	1063	863
0	1072	872
0	1170	970
0	1150	950
0	983	983
0	1163	963
0	1138	938
0	980	980
0	1135	835
1	0	0
0	1248	948
0	1218	918
1	0	0
1	0	0
0	895	595

RawReportData Data

0	1242	942
0	1073	973
1	0	0
0	1185	885
1	0	0
1	0	0
1	0	0
0	1207	907
0	1278	978
1	0	0
1	0	0
0	1285	985
0	1255	955
0	1062	962
0	1258	958
1	0	0

RawReportData Data

1	0	0
1	0	0
1	0	0
1	0	0
0	1243	843
1	0	0
0	773	773
0	1067	667
0	1207	807
1	0	0
0	568	568
0	1283	883
1	0	0
1	0	0
0	590	590
0	945	545

RawReportData Data

1	0	0
1	0	0
1	0	0
1	0	0
0	1082	682
0	1045	845
0	1242	842
0	545	545
1	0	0
1	0	0
0	770	770
0	830	830
0	1348	848
0	802	802
0	867	767
0	1293	793

RawReportData Data

0	1348	848
1	0	0
0	885	785
0	1397	897
0	875	875
0	520	520
0	905	805
0	1287	787
1	0	0
0	800	800
0	812	812
1	0	0
0	1278	778
0	1212	912
0	1360	860
0	663	563

RawReportData Data

0	910	910
0	867	867
0	920	820
0	982	882
0	1387	887
0	828	728
0	1068	868
0	1302	802
0	1362	862
0	813	813
0	1048	848
0	1360	860
0	947	847
0	992	892
0	1022	822
0	1363	863

RawReportData Data

0	797	797
0	885	785
0	1048	948
0	917	917
0	1293	793
0	1322	922
0	1375	875
0	1115	915
0	978	878
0	823	723
0	1135	935
0	1115	915
0	1362	862
1	0	0
0	1137	837
0	1293	793

RawReportData Data

1	0	0
0	973	873
0	978	678
0	1400	900
0	898	698
1	0	0
0	973	673
0	1383	883
0	950	850
1	0	0
1	0	0
0	997	897
0	1312	812
0	1368	868
0	1395	895
0	1105	805

RawReportData Data

0	892	692
1	0	0
1	0	0
1	0	0
1	0	0
1	0	0
1	0	0
1	0	0
1	0	0
1	0	0
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RawReportData Data

1	0	0
1	0	0
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1	0	0
1	0	0
1	0	0
1	0	0
1	0	0
1	0	0

RawReportData Data

0	895	895
1	0	0
1	0	0
1	0	0
1	0	0
1	0	0
1	0	0
1	0	0
1	0	0
1	0	0
1	0	0
1	0	0
1	0	0
1	0	0
1	0	0
1	0	0
1	0	0
0	903	903

RawReportData Data

1	0	0
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Current Total Score (points)	Answer Time (%)
838	32.33%
860	28.00%
953	9.33%
897	20.67%
960	8.00%
905	19.00%
663	67.33%
955	9.00%
940	12.00%
940	12.00%
928	14.33%
922	15.67%
803	39.33%
865	27.00%
765	47.00%

RawReportData Data

932	13.67%
942	11.67%
957	8.67%
933	13.33%
843	31.33%
958	8.33%
962	7.67%
873	25.33%
970	6.00%
1570	73.67%
1743	43.33%
1850	40.67%
1842	31.00%
960	20.67%
905	40.00%
1398	73.00%

RawReportData Data

1848	41.33%
940	18.33%
1798	48.33%
1853	35.00%
922	31.33%
1738	33.00%
1782	36.67%
1712	30.67%
1950	16.33%
1930	22.33%
1964	18.67%
1960	14.67%
1833	22.00%
958	19.00%
1852	42.00%
1781	38.33%

RawReportData Data

970	19.33%
2655	23.00%
2876	13.33%
3027	4.67%
2989	10.67%
1822	27.67%
1860	9.00%
2448	30.00%
2953	19.00%
1912	5.67%
2761	47.33%
3030	4.67%
1895	5.33%
2913	5.00%
2749	46.67%
2652	52.00%

RawReportData Data

3132	3.67%
2993	27.33%
3036	25.67%
3130	6.00%
2983	10.00%
1941	3.33%
3015	7.33%
2919	12.33%
1950	4.00%
3790	33.00%
2876	7.33%
4275	10.33%
4207	16.33%
1822	16.33%
1860	91.67%
3343	81.00%

RawReportData Data

4195	11.67%
2985	5.33%
2761	48.33%
4215	23.00%
1895	4.33%
2913	11.67%
2749	52.33%
3859	18.67%
4410	4.33%
2993	22.00%
3036	26.00%
4415	3.00%
4238	9.00%
3003	7.67%
4273	8.33%
2919	61.67%

RawReportData Data

1950	4.67%
3790	46.67%
2876	47.00%
4275	73.33%
5450	31.33%
1822	45.00%
2633	45.33%
4410	66.67%
5402	38.67%
2985	14.33%
3329	86.33%
5498	23.33%
1895	53.00%
2913	81.33%
3339	82.00%
4804	91.00%

RawReportData Data

4410	26.33%
2993	48.00%
3036	44.33%
4415	17.67%
5320	63.67%
4048	31.00%
5515	31.67%
3464	91.00%
1950	39.33%
3790	30.33%
3646	46.00%
5105	34.00%
6798	30.33%
2624	39.67%
3500	46.67%
5703	41.33%

RawReportData Data

6750	30.33%
2985	46.33%
4214	43.00%
6895	20.67%
2770	25.00%
3433	96.00%
4244	39.00%
6091	42.67%
4410	32.00%
3793	40.00%
3848	37.67%
4415	39.33%
6598	44.33%
5260	17.67%
6875	28.00%
4127	87.33%

RawReportData Data

2860	18.00%
4657	26.67%
4566	36.00%
6087	23.67%
8185	22.67%
3452	54.33%
4568	26.33%
7005	39.67%
8112	27.67%
3798	37.33%
5262	30.33%
8255	28.00%
3717	30.67%
4425	21.67%
5266	35.67%
7454	27.33%

RawReportData Data

5207	40.67%
4678	43.00%
4896	10.33%
5332	16.67%
7891	41.33%
6582	15.67%
8250	25.00%
5242	17.00%
3838	24.33%
5480	55.33%
5701	13.00%
7202	17.00%
9547	27.67%
3452	50.00%
5705	32.67%
8298	41.33%

RawReportData Data

8112	48.67%
4771	25.33%
6240	64.33%
9655	20.00%
4615	60.33%
4425	86.00%
6239	65.33%
8837	23.33%
6157	30.00%
4678	32.33%
4896	30.33%
6329	20.67%
9203	37.67%
7950	26.33%
9645	21.00%
6347	39.00%

RawReportData Data

4730	61.67%
5480	100.00%
5701	100.00%
7202	100.00%
9547	43.67%
3452	100.00%
5705	40.67%
8298	47.67%
8112	8.33%
4771	100.00%
6240	46.33%
9655	30.33%
4615	42.00%
4425	36.33%
6239	51.67%
8837	100.00%

RawReportData Data

6157	56.00%
4678	35.00%
4896	35.33%
6329	19.33%
9203	47.00%
7950	18.67%
9645	33.67%
6347	100.00%
4730	33.33%
5480	100.00%
5701	47.00%
7202	100.00%
9547	36.33%
3452	100.00%
5705	36.67%
8298	100.00%

RawReportData Data

9007	21.00%
4771	23.00%
6240	100.00%
9655	24.00%
4615	44.33%
4425	40.33%
6239	100.00%
8837	100.00%
6157	13.67%
4678	44.67%
4896	42.00%
6329	15.00%
9203	100.00%
7950	18.00%
9645	25.00%
7250	19.33%

RawReportData Data

4730	100.00%
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Answer Time (seconds)
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8,1
14,1

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4
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22,1
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12,2
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6,2
12
21,9

12,4
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11
9,2
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4,4
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5,7
12,6
11,5

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14
15,6

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3
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27,5
24,3

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0,9
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18,5

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11
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6,3
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