


Electronegativity (Block 5)

Played on	12 Nov 2019
Hosted by	JenKrug
Played with	26 players
Played	10 of 10

Overall Performance

Total correct answers (%)	79,23%
Total incorrect answers (%)	20,77%
Average score (points)	9554,6



Feedback

Number of responses	8
How fun was it? (out of 5)	4,00 o
Did you learn something?	75,00%
Do you recommend it?	50,00%
How do you feel?	

Switch tabs/pages to view other result breakdown

Overview

%
%
39 points

ut of 5			
% Yes	25,00% No		
% Yes	50,00% No		
62,50% Positive		12,50% Neutral	

--

Overview

Response	Percentage
Very much	25,00%
Much	25,00%
A little	25,00%
Not at all	25,00%

Electronegativity (Block 5)

Final Scores

Rank	Players
1	mckenna
2	Lindsey
3	mason
4	ok boomer
5	sydney
6	julia
7	jason
8	Camden
9	david
10	Shane
11	Rhys
12	Macon
13	kate
14	sebastian
15	Michael
16	Owen
17	Alec S
18	Liam
19	Max
20	Ashley
21	Ginaa
22	Chelsea

Final Scores

23	maddie
24	Dahlia
25	Bubble Blasters
26	Ok Boomer

Final Scores

Total Score (points)	Correct Answers	Incorrect Answers
13418	10	0
13400	10	0
13329	10	0
13266	10	0
13241	10	0
13093	10	0
11390	9	1
11283	9	1
11265	9	1
11209	9	1
10664	9	1
10613	9	1
10508	9	1
10348	9	1
10333	9	1
10231	9	1
10185	9	1
10110	9	1
9865	8	2
8668	8	2
7006	7	3
6029	6	4

Final Scores

5152	5	5
3816	4	6
0	0	10
0	0	10

Electronegativity (Block 5)

Kahoot! Summary

Rank	Players
1	mckenna
2	Lindsey
3	mason
4	ok boomer
5	sydney
6	julia
7	jason
8	Camden
9	david
10	Shane
11	Rhys
12	Macon
13	kate
14	sebastian
15	Michael

Kahoot! Summary

16	Owen
17	Alec S
18	Liam
19	Max
20	Ashley
21	Ginaa
22	Chelsea
23	maddie
24	Dahlia
25	Bubble Blasters
26	Ok Boomer

Kahoot! Summary

Total Score (points)	Q1
13418	985
13400	984
13329	986
13266	955
13241	948
13093	943
11390	987
11283	966
11265	906
11209	964
10664	938
10613	960
10508	933
10348	946
10333	973

Kahoot! Summary

10231	908
10185	924
10110	907
9865	0
8668	966
7006	931
6029	981
5152	967
3816	887
0	0
0	0

Kahoot! Summary

What is Electronegativity?	Q2
A measure of the tendency of an atom to attract a bonding pair of electrons	1085
A measure of the tendency of an atom to attract a bonding pair of electrons	1092
A measure of the tendency of an atom to attract a bonding pair of electrons	1082
A measure of the tendency of an atom to attract a bonding pair of electrons	1077
A measure of the tendency of an atom to attract a bonding pair of electrons	1092
A measure of the tendency of an atom to attract a bonding pair of electrons	1078
A measure of the tendency of an atom to attract a bonding pair of electrons	1094
A measure of the tendency of an atom to attract a bonding pair of electrons	1069
A measure of the tendency of an atom to attract a bonding pair of electrons	0
A measure of the tendency of an atom to attract a bonding pair of electrons	1074
A measure of the tendency of an atom to attract a bonding pair of electrons	1091
A measure of the tendency of an atom to attract a bonding pair of electrons	1070
A measure of the tendency of an atom to attract a bonding pair of electrons	1058
A measure of the tendency of an atom to attract a bonding pair of electrons	1093
A measure of the tendency of an atom to attract a bonding pair of electrons	1087

Kahoot! Summary

A measure of the tendency of an atom to attract a bonding pair of electrons	1041
A measure of the tendency of an atom to attract a bonding pair of electrons	1054
A measure of the tendency of an atom to attract a bonding pair of electrons	1071
The attraction between an electron and the nucleus in any atom	992
A measure of the tendency of an atom to attract a bonding pair of electrons	1062
A measure of the tendency of an atom to attract a bonding pair of electrons	0
A measure of the tendency of an atom to attract a bonding pair of electrons	1093
A measure of the tendency of an atom to attract a bonding pair of electrons	0
A measure of the tendency of an atom to attract a bonding pair of electrons	0
	0
	0

Kahoot! Summary

Electronegativity is measured using the Pauling Electronegativity Scale	Q3	
	True	1196
	True	1191
	True	1189
	True	1186
	True	1152
	True	1155
	True	1191
	True	1192
		985
	True	1176
	True	1140
	True	0
	True	1134
	True	1184
	True	1173

Kahoot! Summary

	True	1137
	True	1183
	True	1123
	True	1084
	True	1188
		931
	True	0
	False	0
		0
		0
		0

Kahoot! Summary

In which direction does electronegativity trends increase?	Q4
Left to Right	1290
Left to Right	1291
Left to Right	1292
Left to Right	1278
Left to Right	1282
Left to Right	1284
Left to Right	1293
Left to Right	1292
Left to Right	1087
Left to Right	1284
Left to Right	1282
	973
Left to Right	1251
Left to Right	1283
Left to Right	1282

Kahoot! Summary

Left to Right	1287
Left to Right	1280
Left to Right	1269
Left to Right	1188
Left to Right	1288
Left to Right	1091
Down	988
Down	987
Down	986
	0
	0

Kahoot! Summary

What law says that the attraction between protons and electrons increases when they are closer together?	Q5
Coulomb's Law	1383
Coulomb's Law	1379
Coulomb's Law	1385
Coulomb's Law	1386
Coulomb's Law	1388
Coulomb's Law	1375
Coulomb's Law	1395
Coulomb's Law	1364
Coulomb's Law	1175
Coulomb's Law	1363
Coulomb's Law	1339
Coulomb's Law	1072
Coulomb's Law	1337
Coulomb's Law	0
Coulomb's Law	0

Kahoot! Summary

Coulomb's Law	0
Coulomb's Law	0
Coulomb's Law	0
Coulomb's Law	1284
Coulomb's Law	0
Coulomb's Law	0
Coulomb's Law	0
Coulomb's Law	1076
Coulomb's Law	0
	0
	0

Kahoot! Summary

Which has a HIGHER electronegativity?	Q6	
	Nonmetals	1497
	Nonmetals	1483
	Nonmetals	1481
	Nonmetals	1488
	Nonmetals	1466
	Nonmetals	1444
	Nonmetals	1463
	Nonmetals	1489
	Nonmetals	1256
	Nonmetals	1464
	Nonmetals	1441
	Nonmetals	1127
	Nonmetals	1432
	Metals	964
	Metals	984

Kahoot! Summary

Metals	983
Metals	979
Metals	957
Nonmetals	1362
Metals	987
Metals	920
Metals	987
Nonmetals	1172
Metals	936
	0
	0

Kahoot! Summary

Compounds formed between metals and nonmetals generally are?	Q7	
	Ionic	1496
	Ionic	1497
	Ionic	1482
	Ionic	1474
	Ionic	1460
	Ionic	1449
	Ionic	1481
	Ionic	1441
	Ionic	1376
	Ionic	1431
	Ionic	1412
	Ionic	1197
	Ionic	1404
	Ionic	1011
	Ionic	1015

Kahoot! Summary

	Ionic	1024
	Ionic	1002
	Ionic	986
	Ionic	1481
	Ionic	1019
	Ionic	991
	Ionic	0
	Ionic	0
	Ionic	1007
		0
		0

Kahoot! Summary

What is ionic bonding?	Q8
When one atom transfers electrons to another	1497
When one atom transfers electrons to another	1496
When one atom transfers electrons to another	1482
When one atom transfers electrons to another	1489
When one atom transfers electrons to another	1488
When one atom transfers electrons to another	1458
When one atom transfers electrons to another	1493
When one atom transfers electrons to another	1483
When one atom transfers electrons to another	1494
When one atom transfers electrons to another	1474
When one atom transfers electrons to another	0
When one atom transfers electrons to another	1347
When one atom transfers electrons to another	0
When one atom transfers electrons to another	1184
When one atom transfers electrons to another	1177

Kahoot! Summary

When one atom transfers electrons to another	1185
When one atom transfers electrons to another	1166
When one atom transfers electrons to another	1159
When one atom transfers electrons to another	1482
When one atom transfers electrons to another	1172
When one atom transfers electrons to another	1153
A property that is measurable and whose value describes its state	987
A property that is measurable and whose value describes its state	0
When one atom transfers electrons to another	0
	0
	0

Kahoot! Summary

What is covalent bonding?	Q9
When two atoms share electrons	1494
When two atoms share electrons	1493
When two atoms share electrons	1481
When two atoms share electrons	1466
When two atoms share electrons	1487
When two atoms share electrons	1435
When two atoms share electrons	0
When two atoms share electrons	0
When two atoms share electrons	1491
When two atoms share electrons	0
	982
When two atoms share electrons	1447
	972
When two atoms share electrons	1287
When two atoms share electrons	1261

Kahoot! Summary

When two atoms share electrons	1287
When two atoms share electrons	1215
When two atoms share electrons	1254
When two atoms share electrons	0
When two atoms share electrons	0
When two atoms share electrons	0
When two atoms share electrons	0
Occurs when there is an electronegativity difference between bonded atoms	950
	0
	0
	0

Kahoot! Summary

What is polarity?	Q10
The distribution of electric charge around atoms or chemical groups	1495
The distribution of electric charge around atoms or chemical groups	1494
The distribution of electric charge around atoms or chemical groups	1469
The distribution of electric charge around atoms or chemical groups	1467
The distribution of electric charge around atoms or chemical groups	1478
The distribution of electric charge around atoms or chemical groups	1472
Elements with great differences in electronegativity	993
Elements with great differences in electronegativity	987
The distribution of electric charge around atoms or chemical groups	1495
Elements with great differences in electronegativity	979
The distribution of electric charge around atoms or chemical groups	1039
The distribution of electric charge around atoms or chemical groups	1420
The distribution of electric charge around atoms or chemical groups	987
The distribution of electric charge around atoms or chemical groups	1396
The distribution of electric charge around atoms or chemical groups	1381

Kahoot! Summary

The distribution of electric charge around atoms or chemical groups	1379
The distribution of electric charge around atoms or chemical groups	1382
The distribution of electric charge around atoms or chemical groups	1384
Elements with great differences in electronegativity	992
Elements with great differences in electronegativity	986
Elements with great differences in electronegativity	989
Elements with great differences in electronegativity	993
The distribution of electric charge around atoms or chemical groups	0
	0
	0
	0

Kahoot! Summary

Polarity is not measured by difference in electronegativity	
	False
	False
	False
	False
	False
	False
	False
	False
	False
	False
	False
	False
	False
	False

Kahoot! Summary

False
False
False
False
False
False
False
True

Electrone
1 Quiz
Correct answers
Players correct (
Question duratic
Answer Sum
Answer options
Is answer correc
Number of answ
Average time tal
Answer Deta
Players
Alec S
Ashley
Bubble Blasters
Camden
Chelsea
Dahlia
Ginaa
Liam
Lindsey
Macon

1 Quiz

Max
Michael
Ok Boomer
Owen
Rhys
Shane
david
jason
julia
kate
maddie
mason
mckenna
ok boomer
sebastian
sydney

Electronegativity (Block 5)	
What is Electronegativity?	
is	A measu
(%)	88,46%
on	90 secur

Summary	
	▲
st?	
ers received	
ken to answer (seconds)	




Details	
	Answer
	✓
	✓
	✗
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓

1 Quiz

	X
	✓
	X
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓

1 Quiz

ire of the tendency of an atom to attract a bonding pair of elect
nds

A measure of the tendency of an atom to attract a bonding pair of electrons	
	
23	
9,04	

	Score (p
A measure of the tendency of an atom to attract a bonding pair of electrons	924
A measure of the tendency of an atom to attract a bonding pair of electrons	966
	0
A measure of the tendency of an atom to attract a bonding pair of electrons	966
A measure of the tendency of an atom to attract a bonding pair of electrons	981
A measure of the tendency of an atom to attract a bonding pair of electrons	887
A measure of the tendency of an atom to attract a bonding pair of electrons	931
A measure of the tendency of an atom to attract a bonding pair of electrons	907
A measure of the tendency of an atom to attract a bonding pair of electrons	984
A measure of the tendency of an atom to attract a bonding pair of electrons	960

1 Quiz

The attraction between an electron and the nucleus in any atom	0
A measure of the tendency of an atom to attract a bonding pair of electrons	973
	0
A measure of the tendency of an atom to attract a bonding pair of electrons	908
A measure of the tendency of an atom to attract a bonding pair of electrons	938
A measure of the tendency of an atom to attract a bonding pair of electrons	964
A measure of the tendency of an atom to attract a bonding pair of electrons	906
A measure of the tendency of an atom to attract a bonding pair of electrons	987
A measure of the tendency of an atom to attract a bonding pair of electrons	943
A measure of the tendency of an atom to attract a bonding pair of electrons	933
A measure of the tendency of an atom to attract a bonding pair of electrons	967
A measure of the tendency of an atom to attract a bonding pair of electrons	986
A measure of the tendency of an atom to attract a bonding pair of electrons	985
A measure of the tendency of an atom to attract a bonding pair of electrons	955
A measure of the tendency of an atom to attract a bonding pair of electrons	946
A measure of the tendency of an atom to attract a bonding pair of electrons	948

trons

The complete transfer of valence electrons between atoms	<div><div></div></div>
X	
0	
0,00	

(points)	Current
	924
	966
	0
	966
	981
	887
	931
	907
	984
	960

1 Quiz

	0
	973
	0
	908
	938
	964
	906
	987
	943
	933
	967
	986
	985
	955
	946
	948

1 Quiz

The attraction between an electron and the nucleus in anv atom	<input type="checkbox"/>
X	
1	
2,10	

Total Score (points)	Answer ti
	13,7
	6,1
	90
	6,2
	3,4
	20,3
	12,5
	16,8
	2,9
	7,2

1 Quiz

	2,1
	4,8
	90
	16,6
	11,1
	6,4
	16,9
	2,4
	10,3
	12,1
	5,9
	2,5
	2,7
	8,1
	9,7
	9,3

A process where one or more substances are altered into new substances

Rate of reaction

Time (seconds)

0

0,00

X

1 Quiz

Electrone
2 Quiz
Correct answers
Players correct (
Question duratic
Answer Sum
Answer options
Is answer correc
Number of answ
Average time tal
Answer Deta
Players
Alec S
Ashley
Bubble Blasters
Camden
Chelsea
Dahlia
Ginaa
Liam
Lindsey
Macon

2 Quiz

Max
Michael
Ok Boomer
Owen
Rhys
Shane
david
jason
julia
kate
maddie
mason
mckenna
ok boomer
sebastian
sydney

Electronegativity (Block 5)	
Electronegativity is measured using the Pauling Electronegativity	
is	True
(%)	76,92%
on	90 secor



Summary	
	▲
st?	
ers received	
ken to answer (seconds)	

ails	
	Answer
	✓
	✓
	✗
	✓
	✓
	✗
	✗
	✓
	✓
	✓

2 Quiz

	✓
	✓
	✗
	✓
	✓
	✓
	✗
	✓
	✓
	✓
	✗
	✓
	✓
	✓
	✓
	✓

Scale
nds

False	
X	
1	
4,40	

	Score (p
True	1054
True	1062
	0
True	1069
True	1093
	0
	0
True	1071
True	1092
True	1070

2 Quiz

True	992
True	1087
	0
True	1041
True	1091
True	1074
	0
True	1094
True	1078
True	1058
False	0
True	1082
True	1085
True	1077
True	1093
True	1092

2 Quiz

True	<div><div></div></div>
<div>✓</div>	
20	
4,02	

oints)	Current
	1978
	2028
	0
	2035
	2074
	887
	931
	1978
	2076
	2030

2 Quiz

	992
	2060
	0
	1949
	2029
	2038
	906
	2081
	2021
	1991
	967
	2068
	2070
	2032
	2039
	2040

2 Quiz

Total Score (points)	Answer ti
	8,2
	6,9
	90
	5,5
	1,3
	90
	90
	5,2
	1,4
	5,4

2 Quiz

	1,4
	2,3
	90
	10,7
	1,7
	4,7
	90
	1,1
	4
	7,6
	4,4
	3,3
	2,7
	4,1
	1,3
	1,5

2 Quiz

[illegible]

2 Quiz

Electrone
3 Quiz
Correct answers
Players correct (
Question duratic
Answer Sum
Answer options
Is answer correc
Number of answ
Average time tal
Answer Deta
Players
Alec S
Ashley
Bubble Blasters
Camden
Chelsea
Dahlia
Ginaa
Liam
Lindsey
Macon

3 Quiz

Max
Michael
Ok Boomer
Owen
Rhys
Shane
david
jason
julia
kate
maddie
mason
mckenna
ok boomer
sebastian
sydney

Electronegativity (Block 5)	
In which direction does electronegativity trends increase?	
Direction	Left to Right
Score (%)	76,92%
Time taken	90 seconds

Summary	
Correct answers	4
Wrong answers	1
Questions received	5
Time taken to answer (seconds)	90



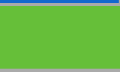
Details	
Question	Answer
1. In which direction does electronegativity trends increase?	Left to Right
2. Electronegativity increases from left to right across the periodic table.	True
3. Electronegativity increases from top to bottom across the periodic table.	False
4. Electronegativity increases from left to right across the periodic table.	True
5. Electronegativity increases from top to bottom across the periodic table.	False
6. Electronegativity increases from left to right across the periodic table.	True
7. Electronegativity increases from top to bottom across the periodic table.	False
8. Electronegativity increases from left to right across the periodic table.	True
9. Electronegativity increases from top to bottom across the periodic table.	False
10. Electronegativity increases from left to right across the periodic table.	True
11. Electronegativity increases from top to bottom across the periodic table.	False
12. Electronegativity increases from left to right across the periodic table.	True
13. Electronegativity increases from top to bottom across the periodic table.	False
14. Electronegativity increases from left to right across the periodic table.	True
15. Electronegativity increases from top to bottom across the periodic table.	False

3 Quiz

	✓
	✓
	✗
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✗
	✓
	✓
	✓
	✓
	✓

3 Quiz

ight
nds

Down	
	
	3
	13,03

	Score (p
Left to Right	1183
Left to Right	1188
	0
Left to Right	1192
Down	0
Down	0
Left to Right	931
Left to Right	1123
Left to Right	1191
	0

3 Quiz

Left to Right	1084
Left to Right	1173
	0
Left to Right	1137
Left to Right	1140
Left to Right	1176
Left to Right	985
Left to Right	1191
Left to Right	1155
Left to Right	1134
Down	0
Left to Right	1189
Left to Right	1196
Left to Right	1186
Left to Right	1184
Left to Right	1152

3 Quiz

Left to Right	
✓	
20	
5,49	

oints)	Current
	3161
	3216
	0
	3227
	2074
	887
	1862
	3101
	3267
	2030

3 Quiz

	2076
	3233
	0
	3086
	3169
	3214
	1891
	3272
	3176
	3125
	967
	3257
	3266
	3218
	3223
	3192

3 Quiz

Total Score (points)	Answer t
	3,1
	2,1
	90
	1,4
	7,4
	25,5
	12,5
	13,9
	1,7
	90

3 Quiz

	2,9
	4,9
	90
	11,4
	10,8
	4,3
	2,7
	1,6
	8,1
	11,8
	6,2
	2
	0,7
	2,5
	2,8
	8,6

3 Quiz

[illegible]

3 Quiz

Electronics
4 Quiz
Correct answers
Players correct (0)
Question duration
Answer Summary
Answer options
Is answer correct
Number of answers
Average time taken
Answer Details
Players
Alec S
Ashley
Bubble Blasters
Camden
Chelsea
Dahlia
Ginaa
Liam
Lindsey
Macon

4 Quiz

Max
Michael
Ok Boomer
Owen
Rhys
Shane
david
jason
julia
kate
maddie
mason
mckenna
ok boomer
sebastian
sydney

Negativity (Block 5)	
What law says that the attraction between protons and electrons is	
s	Coulomb's
(%)	92,31%
on	90 seconds

Summary	
	▲
st?	
ers received	
ken to answer (seconds)	

Details	
	Answer
	✓
	✓
	✗
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓

4 Quiz

	✓
	✓
	✗
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓

increases when they are closer together?
Boyle's Law
ends

Boyle's Law	◆
X	
0	
0,00	

	Score (p
Coulomb's Law	1280
Coulomb's Law	1288
	0
Coulomb's Law	1292
Coulomb's Law	988
Coulomb's Law	986
Coulomb's Law	1091
Coulomb's Law	1269
Coulomb's Law	1291
Coulomb's Law	973

4 Quiz

Coulomb's Law	1188
Coulomb's Law	1282
	0
Coulomb's Law	1287
Coulomb's Law	1282
Coulomb's Law	1284
Coulomb's Law	1087
Coulomb's Law	1293
Coulomb's Law	1284
Coulomb's Law	1251
Coulomb's Law	987
Coulomb's Law	1292
Coulomb's Law	1290
Coulomb's Law	1278
Coulomb's Law	1283
Coulomb's Law	1282

4 Quiz

Aristotle's Law	
X	
0	
0,00	

oints)	Current
	4441
	4504
	0
	4519
	3062
	1873
	2953
	4370
	4558
	3003

4 Quiz

	3264
	4515
	0
	4373
	4451
	4498
	2978
	4565
	4460
	4376
	1954
	4549
	4556
	4496
	4506
	4474

4 Quiz

Coulomb's Law	<input type="checkbox"/>
<input checked="" type="checkbox"/>	
24	
2,95	

Total Score (points)	Answer t
	3,6
	2,2
	90
	1,4
	2,1
	2,6
	1,6
	5,6
	1,6
	4,9

4 Quiz

	2,1
	3,3
	90
	2,4
	3,2
	2,8
	2,4
	1,3
	2,9
	8,9
	2,4
	1,4
	1,8
	4
	3,1
	3,3

[illegible]

4 Quiz

Electrone
5 Quiz
Correct answers
Players correct (
Question duratic
Answer Sum
Answer options
Is answer correc
Number of answ
Average time tal
Answer Deta
Players
Alec S
Ashley
Bubble Blasters
Camden
Chelsea
Dahlia
Ginaa
Liam
Lindsey
Macon

5 Quiz

Max
Michael
Ok Boomer
Owen
Rhys
Shane
david
jason
julia
kate
maddie
mason
mckenna
ok boomer
sebastian
sydney

Electronegativity (Block 5)	
Which has a HIGHER electronegativity?	
Nonmetals	Nonmetals
57,69%	57,69%
90 seconds	90 seconds

Summary	
	▲
Correct?	
Answers received	
Time taken to answer (seconds)	



Details	
	Answer
	X
	X
	X
	✓
	X
	X
	X
	X
	✓
	✓

5 Quiz

	✓
	✗
	✗
	✗
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✗
	✓

5 Quiz

als
nds

Metals	
X	
9	
5,30	

	Score (p
Metals	0
Metals	0
	0
Nonmetals	1364
Metals	0
Metals	0
Metals	0
Metals	0
Nonmetals	1379
Nonmetals	1072

5 Quiz

Nonmetals	1284
Metals	0
	0
Metals	0
Nonmetals	1339
Nonmetals	1363
Nonmetals	1175
Nonmetals	1395
Nonmetals	1375
Nonmetals	1337
Nonmetals	1076
Nonmetals	1385
Nonmetals	1383
Nonmetals	1386
Metals	0
Nonmetals	1388

5 Quiz

Nonmetals	
✓	
15	
4,79	

oints)	Current
	4441
	4504
	0
	5883
	3062
	1873
	2953
	4370
	5937
	4075

5 Quiz

	4548
	4515
	0
	4373
	5790
	5861
	4153
	5960
	5835
	5713
	3030
	5934
	5939
	5882
	4506
	5862

5 Quiz

Total Score (points)	Answer ti
	3,4
	2,7
	90
	6,5
	2,1
	4,5
	4
	5,8
	3,7
	5,1

5 Quiz

	2,8
	7,3
	90
	13,8
	11
	6,7
	4,5
	0,9
	4,5
	11,3
	4,4
	2,7
	3
	2,5
	4,1
	2,2

5 Quiz

[illegible]

5 Quiz

Electrone
6 Quiz
Correct answers
Players correct (
Question duratic
Answer Sum
Answer options
Is answer correc
Number of answ
Average time tal
Answer Deta
Players
Alec S
Ashley
Bubble Blasters
Camden
Chelsea
Dahlia
Ginaa
Liam
Lindsey
Macon

6 Quiz

Max
Michael
Ok Boomer
Owen
Rhys
Shane
david
jason
julia
kate
maddie
mason
mckenna
ok boomer
sebastian
sydney

Negativity (Block 5)	
Compounds formed between metals and nonmetals generally are	
s	Ionic
(%)	92,31%
on	90 secur

Summary	
	▲
st?	
ers received	
ken to answer (seconds)	

ails	
	Answer
	✓
	✓
	✗
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓

6 Quiz

	✓
	✓
	✗
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓

6 Quiz

?	
nds	

Atomic	◆
X	
0	
0,00	

	Score (p
Ionic	979
Ionic	987
	0
Ionic	1489
Ionic	987
Ionic	936
Ionic	920
Ionic	957
Ionic	1483
Ionic	1127

6 Quiz

Ionic	1362
Ionic	984
	0
Ionic	983
Ionic	1441
Ionic	1464
Ionic	1256
Ionic	1463
Ionic	1444
Ionic	1432
Ionic	1172
Ionic	1481
Ionic	1497
Ionic	1488
Ionic	964
Ionic	1466

6 Quiz

Chemical	
X	
0	
0,00	

oints)	Current
	5420
	5491
	0
	7372
	4049
	2809
	3873
	5327
	7420
	5202

6 Quiz

	5910
	5499
	0
	5356
	7231
	7325
	5409
	7423
	7279
	7145
	4202
	7415
	7436
	7370
	5470
	7328

6 Quiz

Metallic	<input checked="" type="checkbox"/>
X	<input type="checkbox"/>
0	
0,00	

Total Score (points)	Answer t
	3,7
	2,4
	90
	2
	2,4
	11,6
	14,4
	7,8
	3
	13,2

6 Quiz

	6,9
	2,8
	90
	3,1
	10,6
	6,4
	8
	6,7
	10,1
	12,3
	5
	3,4
	0,5
	2,2
	6,4
	6,2

6 Quiz

Time (seconds)	Value
0	0
1	1
2	2
3	3
4	4
5	5
6	6
7	7
8	8
9	9
10	10
11	11
12	12
13	13
14	14
15	15
16	16
17	17
18	18
19	19
20	20
21	21
22	22
23	23
24	24
25	25
26	26
27	27
28	28
29	29
30	30
31	31
32	32
33	33
34	34
35	35
36	36
37	37
38	38
39	39
40	40
41	41
42	42
43	43
44	44
45	45
46	46
47	47
48	48
49	49
50	50
51	51
52	52
53	53
54	54
55	55
56	56
57	57
58	58
59	59
60	60
61	61
62	62
63	63
64	64
65	65
66	66
67	67
68	68
69	69
70	70
71	71
72	72
73	73
74	74
75	75
76	76
77	77
78	78
79	79
80	80
81	81
82	82
83	83
84	84
85	85
86	86
87	87
88	88
89	89
90	90
91	91
92	92
93	93
94	94
95	95
96	96
97	97
98	98
99	99
100	100

6 Quiz

Electronics
7 Quiz
Correct answers
Players correct (0)
Question duration
Answer Summary
Answer options
Is answer correct
Number of answers
Average time taken
Answer Details
Players
Alec S
Ashley
Bubble Blasters
Camden
Chelsea
Dahlia
Ginaa
Liam
Lindsey
Macon

7 Quiz

Max
Michael
Ok Boomer
Owen
Rhys
Shane
david
jason
julia
kate
maddie
mason
mckenna
ok boomer
sebastian
sydney

Negativity (Block 5)	
What is ionic bonding?	
s	When or
(%)	84,62%
on	90 secur

Summary	
	▲
st?	
ers received	
ken to answer (seconds)	

ails	
	Answer
	✓
	✓
	✗
	✓
	✗
	✓
	✓
	✓
	✓
	✓

7 Quiz

	✓
	✓
	✗
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✗
	✓
	✓
	✓
	✓
	✓

ne atom transfers electrons to another
nds

Any of a material's propertie that becomes evident during a reaction	◆
X	
0	
0,00	

	Score (p
When one atom transfers electrons to another	1002
When one atom transfers electrons to another	1019
	0
When one atom transfers electrons to another	1441
A property that is measurable and whose value describes its state	0
When one atom transfers electrons to another	1007
When one atom transfers electrons to another	991
When one atom transfers electrons to another	986
When one atom transfers electrons to another	1497
When one atom transfers electrons to another	1197

7 Quiz

When one atom transfers electrons to another	1481
When one atom transfers electrons to another	1015
	0
When one atom transfers electrons to another	1024
When one atom transfers electrons to another	1412
When one atom transfers electrons to another	1431
When one atom transfers electrons to another	1376
When one atom transfers electrons to another	1481
When one atom transfers electrons to another	1449
When one atom transfers electrons to another	1404
A property that is measurable and whose value describes its state	0
When one atom transfers electrons to another	1482
When one atom transfers electrons to another	1496
When one atom transfers electrons to another	1474
When one atom transfers electrons to another	1011
When one atom transfers electrons to another	1460

7 Quiz

A property that is measurable and whose value describes its state	<div></div>
X	<div></div>
2	
8,35	

oints)	Current
	6422
	6510
	0
	8813
	4049
	3816
	4864
	6313
	8917
	6399

7 Quiz

	7391
	6514
	0
	6380
	8643
	8756
	6785
	8904
	8728
	8549
	4202
	8897
	8932
	8844
	6481
	8788

7 Quiz

When one atom transfers electrons to another	<input type="checkbox"/>
<input checked="" type="checkbox"/>	
22	
11,17	

Total Score (points)	Answer ti
	17,7
	14,6
	90
	10,6
	5,1
	16,7
	19,7
	20,5
	0,6
	18,5

7 Quiz

	3,5
	15,3
	90
	13,7
	15,9
	12,4
	4,4
	3,5
	9,2
	17,2
	11,6
	3,2
	0,7
	4,6
	16,1
	7,2

7 Quiz

[illegible]

7 Quiz

Electrone
8 Quiz
Correct answers
Players correct (
Question duratic
Answer Sum
Answer options
Is answer correc
Number of answ
Average time tal
Answer Deta
Players
Alec S
Ashley
Bubble Blasters
Camden
Chelsea
Dahlia
Ginaa
Liam
Lindsey
Macon

8 Quiz

Max
Michael
Ok Boomer
Owen
Rhys
Shane
david
jason
julia
kate
maddie
mason
mckenna
ok boomer
sebastian
sydney

Negativity (Block 5)	
What is covalent bonding?	
s	When tw
(%)	76,92%
on	90 secur

Summary	
	▲
st?	
ers received	
ken to answer (seconds)	

ails	
	Answer
	✓
	✓
	✗
	✓
	✓
	✗
	✓
	✓
	✓
	✓

8 Quiz

	✓
	✓
	✗
	✓
	✗
	✓
	✓
	✓
	✓
	✗
	✗
	✓
	✓
	✓
	✓
	✓

8 Quiz

o atoms share electrons
nds

Distribution of electric charge around atoms	◆
X	
0	
0,00	

	Score (p
When two atoms share electrons	1166
When two atoms share electrons	1172
	0
When two atoms share electrons	1483
When two atoms share electrons	987
	0
When two atoms share electrons	1153
When two atoms share electrons	1159
When two atoms share electrons	1496
When two atoms share electrons	1347

8 Quiz

When two atoms share electrons	1482
When two atoms share electrons	1177
	0
When two atoms share electrons	1185
	0
When two atoms share electrons	1474
When two atoms share electrons	1494
When two atoms share electrons	1493
When two atoms share electrons	1458
	0
Occurs when there is an electronegativity difference between bonded atoms	0
When two atoms share electrons	1482
When two atoms share electrons	1497
When two atoms share electrons	1489
When two atoms share electrons	1184
When two atoms share electrons	1488

8 Quiz

When two atoms share electrons	<div></div>
<div>✓</div>	
20	
3,89	

oints)	Current
	7588
	7682
	0
	10296
	5036
	3816
	6017
	7472
	10413
	7746

8 Quiz

	8873
	7691
	0
	7565
	8643
	10230
	8279
	10397
	10186
	8549
	4202
	10379
	10429
	10333
	7665
	10276

8 Quiz

When one atom transfers electrons to another	<input checked="" type="checkbox"/>
X	
0	
0,00	

Total Score (points)	Answer t
	6,1
	5
	90
	3
	2,3
	90
	8,4
	7,3
	0,7
	9,5

8 Quiz

	3,3
	4,1
	90
	2,7
	90
	4,7
	1
	1,3
	7,5
	90
	6,3
	3,2
	0,5
	2
	2,9
	2,2

8 Quiz

	time (seconds)
Occurs when there is an electronegativity difference between bonded atoms	X
	6,30
	1

8 Quiz

Electrone
9 Quiz
Correct answers
Players correct (
Question duratic
Answer Sum
Answer options
Is answer correc
Number of answ
Average time tal
Answer Deta
Players
Alec S
Ashley
Bubble Blasters
Camden
Chelsea
Dahlia
Ginaa
Liam
Lindsey
Macon

Max
Michael
Ok Boomer
Owen
Rhys
Shane
david
jason
julia
kate
maddie
mason
mckenna
ok boomer
sebastian
sydney

Negativity (Block 5)	
What is polarity?	
s	The distr
(%)	61,54%
on	90 secur

Summary	
	▲
st?	
ers received	
ken to answer (seconds)	

ails	
	Answer
	✓
	X
	X
	X
	X
	X
	X
	✓
	✓
	✓

9 Quiz

	X
	✓
	X
	✓
	✓
	X
	✓
	X
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓

tribution of electric charge around atoms or chemical groups
nds




Elements with similar electronegativites	◆
X	
0	
0,00	

	Score (p
The distribution of electric charge around atoms or chemical groups	1215
Elements with great differences in electronegativity	0
	0
Elements with great differences in electronegativity	0
Elements with great differences in electronegativity	0
	0
Elements with great differences in electronegativity	0
The distribution of electric charge around atoms or chemical groups	1254
The distribution of electric charge around atoms or chemical groups	1493
The distribution of electric charge around atoms or chemical groups	1447

9 Quiz

Elements with great differences in electronegativity	0
The distribution of electric charge around atoms or chemical groups	1261
	0
The distribution of electric charge around atoms or chemical groups	1287
The distribution of electric charge around atoms or chemical groups	982
Elements with great differences in electronegativity	0
The distribution of electric charge around atoms or chemical groups	1491
Elements with great differences in electronegativity	0
The distribution of electric charge around atoms or chemical groups	1435
The distribution of electric charge around atoms or chemical groups	972
The distribution of electric charge around atoms or chemical groups	950
The distribution of electric charge around atoms or chemical groups	1481
The distribution of electric charge around atoms or chemical groups	1494
The distribution of electric charge around atoms or chemical groups	1466
The distribution of electric charge around atoms or chemical groups	1287
The distribution of electric charge around atoms or chemical groups	1487

9 Quiz

The distribution of electric charge around atoms or chemical groups	
	
16	
5,61	

oints)	Current
	8803
	7682
	0
	10296
	5036
	3816
	6017
	8726
	11906
	9193

9 Quiz

	8873
	8952
	0
	8852
	9625
	10230
	9770
	10397
	11621
	9521
	5152
	11860
	11923
	11799
	8952
	11763

9 Quiz

Elements with great differences in electronegativity	<input type="checkbox"/>
X	
7	
8,69	

Total Score (points)	Answer ti
	15,3
	16,9
	90
	5,8
	6,5
	90
	18
	8,2
	1,3
	9,5

9 Quiz

	3,5
	7,1
	90
	2,4
	3,2
	8
	1,6
	2,1
	11,7
	5
	9
	3,5
	1,1
	6,1
	2,4
	2,4

ime (seconds)

A group with the smallest value

X

0

0,00

9 Quiz

Electrone
10 Quiz
Correct answers
Players correct (
Question duratic
Answer Sum
Answer options
Is answer correc
Number of answ
Average time tal
Answer Deta
Players
Alec S
Ashley
Bubble Blasters
Camden
Chelsea
Dahlia
Ginaa
Liam
Lindsey
Macon

10 Quiz

Max
Michael
Ok Boomer
Owen
Rhys
Shane
david
jason
julia
kate
maddie
mason
mckenna
ok boomer
sebastian
sydney

Electronegativity (Block 5)	
Polarity is not measured by difference in electronegativity	
is	False
(%)	84,62%
on	90 seconds

Summary	
	▲
st?	
ers received	
ken to answer (seconds)	




Details	
	Answer
	✓
	✓
	✗
	✓
	✓
	✗
	✓
	✓
	✓
	✓

10 Quiz

	✓
	✓
	✗
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✓
	✗
	✓
	✓
	✓
	✓
	✓

10 Quiz

nds

False	
	
22	
4,43	

	Score (p
False	1382
False	986
	0
False	987
False	993
	0
False	989
False	1384
False	1494
False	1420

10 Quiz

False	992
False	1381
	0
False	1379
False	1039
False	979
False	1495
False	993
False	1472
False	987
True	0
False	1469
False	1495
False	1467
False	1396
False	1478

10 Quiz

True	<input checked="" type="radio"/>
X	
1	
5,70	

oints)	Current
	10185
	8668
	0
	11283
	6029
	3816
	7006
	10110
	13400
	10613

10 Quiz

	9865
	10333
	0
	10231
	10664
	11209
	11265
	11390
	13093
	10508
	5152
	13329
	13418
	13266
	10348
	13241

10 Quiz

Total Score (points)	Answer t
	3,3
	2,6
	90
	2,4
	1,2
	90
	2
	2,8
	1,1
	14,4

10 Quiz

	1,4
	3,5
	90
	3,7
	11
	3,7
	0,9
	1,2
	5
	20,4
	5,7
	5,5
	0,9
	5,9
	0,7
	3,9

10 Quiz

Question Number
1 Quiz
1 Quiz
1 Quiz
1 Quiz
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1 Quiz
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2 Quiz
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Question
What is Electronegativity?
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Electronegativity is measured using the Pauling Electronegativity Scale
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What law says that the attraction between protons and electrons increases when they are closer together?

What law says that the attraction between protons and electrons increases when they are closer together?
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Compounds formed between metals and nonmetals generally are?
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Polarity is not measured by difference in electronegativity
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Polarity is not measured by difference in electronegativity

Answer 1	Answer 2
A measure of the tendency of an atom to attract a bonding pair of electrons	The complete transfer of valence electrons between atoms
A measure of the tendency of an atom to attract a bonding pair of electrons	The complete transfer of valence electrons between atoms
A measure of the tendency of an atom to attract a bonding pair of electrons	The complete transfer of valence electrons between atoms
A measure of the tendency of an atom to attract a bonding pair of electrons	The complete transfer of valence electrons between atoms
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A measure of the tendency of an atom to attract a bonding pair of electrons	The complete transfer of valence electrons between atoms
False	True
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False	True

RawReportData Data

False	True
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Boyle's Law	Aristotle's Law

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Metals	Nonmetals
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Metals	Nonmetals
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Metals	Nonmetals

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Metals	Nonmetals
Metals	Nonmetals
Metals	Nonmetals
Atomic	Chemical
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Atomic	Chemical
Atomic	Chemical
Any of a material's propertie that becomes evident during a reaction	A property that is measurable and whose value describes its state
Any of a material's propertie that becomes evident during a reaction	A property that is measurable and whose value describes its state
Any of a material's propertie that becomes evident during a reaction	A property that is measurable and whose value describes its state

Any of a material's propertie that becomes evident during a reaction	A property that is measurable and whose value describes its state
Any of a material's propertie that becomes evident during a reaction	A property that is measurable and whose value describes its state
Any of a material's propertie that becomes evident during a reaction	A property that is measurable and whose value describes its state
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Any of a material's propertie that becomes evident during a reaction	A property that is measurable and whose value describes its state
Any of a material's propertie that becomes evident during a reaction	A property that is measurable and whose value describes its state
Any of a material's propertie that becomes evident during a reaction	A property that is measurable and whose value describes its state
Any of a material's propertie that becomes evident during a reaction	A property that is measurable and whose value describes its state
Distribution of electric charge around atoms	When two atoms share electrons
Distribution of electric charge around atoms	When two atoms share electrons
Distribution of electric charge around atoms	When two atoms share electrons
Distribution of electric charge around atoms	When two atoms share electrons
Distribution of electric charge around atoms	When two atoms share electrons
Distribution of electric charge around atoms	When two atoms share electrons
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Distribution of electric charge around atoms	When two atoms share electrons
Distribution of electric charge around atoms	When two atoms share electrons
Distribution of electric charge around atoms	When two atoms share electrons

Distribution of electric charge around atoms	When two atoms share electrons
Elements with similar electronegativites	The distribution of electric charge around atoms or chemical groups
Elements with similar electronegativites	The distribution of electric charge around atoms or chemical groups
Elements with similar electronegativites	The distribution of electric charge around atoms or chemical groups
Elements with similar electronegativites	The distribution of electric charge around atoms or chemical groups
Elements with similar electronegativites	The distribution of electric charge around atoms or chemical groups
Elements with similar electronegativites	The distribution of electric charge around atoms or chemical groups
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Elements with similar electronegativites	The distribution of electric charge around atoms or chemical groups
False	True
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RawReportData Data

False	True
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False	True
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RawReportData Data

False	True
False	True
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False	True
False	True

Answer 3	Answer 4
The attraction between an electron and the nucleus in any atom	A process where one or more substances are altered into new substances
The attraction between an electron and the nucleus in any atom	A process where one or more substances are altered into new substances
The attraction between an electron and the nucleus in any atom	A process where one or more substances are altered into new substances
The attraction between an electron and the nucleus in any atom	A process where one or more substances are altered into new substances
The attraction between an electron and the nucleus in any atom	A process where one or more substances are altered into new substances
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The attraction between an electron and the nucleus in any atom	A process where one or more substances are altered into new substances
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The attraction between an electron and the nucleus in any atom	A process where one or more substances are altered into new substances
The attraction between an electron and the nucleus in any atom	A process where one or more substances are altered into new substances
The attraction between an electron and the nucleus in any atom	A process where one or more substances are altered into new substances

Coulomb's Law	Democritus' Law

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Metallic	Ionic
Metallic	Ionic
When one atom transfers electrons to another	
When one atom transfers electrons to another	
When one atom transfers electrons to another	

When one atom transfers electrons to another	
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When one atom transfers electrons to another	
When one atom transfers electrons to another	
When one atom transfers electrons to another	Occurs when there is an electronegativity difference between bonded atoms
When one atom transfers electrons to another	Occurs when there is an electronegativity difference between bonded atoms
When one atom transfers electrons to another	Occurs when there is an electronegativity difference between bonded atoms
When one atom transfers electrons to another	Occurs when there is an electronegativity difference between bonded atoms
When one atom transfers electrons to another	Occurs when there is an electronegativity difference between bonded atoms
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When one atom transfers electrons to another	Occurs when there is an electronegativity difference between bonded atoms
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When one atom transfers electrons to another	Occurs when there is an electronegativity difference between bonded atoms

When one atom transfers electrons to another	Occurs when there is an electronegativity difference between bonded atoms
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When one atom transfers electrons to another	Occurs when there is an electronegativity difference between bonded atoms
When one atom transfers electrons to another	Occurs when there is an electronegativity difference between bonded atoms

When one atom transfers electrons to another	Occurs when there is an electronegativity difference between bonded atoms
Elements with great differences in electronegativity	A group with the smallest value
Elements with great differences in electronegativity	A group with the smallest value
Elements with great differences in electronegativity	A group with the smallest value
Elements with great differences in electronegativity	A group with the smallest value
Elements with great differences in electronegativity	A group with the smallest value
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Elements with great differences in electronegativity	A group with the smallest value

Elements with great differences in electronegativity	A group with the smallest value
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Elements with great differences in electronegativity	A group with the smallest value
Elements with great differences in electronegativity	A group with the smallest value
Elements with great differences in electronegativity	A group with the smallest value
Elements with great differences in electronegativity	A group with the smallest value

RawReportData Data

Correct Answers	Time Allotted to Answer (seconds)
A measure of the tendency of an atom to attract a bonding pair of electrons	90
A measure of the tendency of an atom to attract a bonding pair of electrons	90
A measure of the tendency of an atom to attract a bonding pair of electrons	90
A measure of the tendency of an atom to attract a bonding pair of electrons	90
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A measure of the tendency of an atom to attract a bonding pair of electrons	90
A measure of the tendency of an atom to attract a bonding pair of electrons	90
True	90
True	90
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True	90

RawReportData Data

True	90
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RawReportData Data

True	90
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Left to Right	90
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Coulomb's Law	90

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When one atom transfers electrons to another	90
When one atom transfers electrons to another	90
When one atom transfers electrons to another	90

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When two atoms share electrons	90
The distribution of electric charge around atoms or chemical groups	90
The distribution of electric charge around atoms or chemical groups	90
The distribution of electric charge around atoms or chemical groups	90
The distribution of electric charge around atoms or chemical groups	90
The distribution of electric charge around atoms or chemical groups	90
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The distribution of electric charge around atoms or chemical groups	90
False	90
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RawReportData Data

False	90
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RawReportData Data

False	90
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Players
Alec S
Ashley
Bubble Blasters
Camden
Chelsea
Dahlia
Ginaa
Liam
Lindsey
Macon
Max
Michael
Ok Boomer
Owen
Rhys

Shane
david
jason
julia
kate
maddie
mason
mckenna
ok boomer
sebastian
sydney
Alec S
Ashley
Bubble Blasters
Camden
Chelsea

Dahlia
Ginaa
Liam
Lindsey
Macon
Max
Michael
Ok Boomer
Owen
Rhys
Shane
david
jason
julia
kate
maddie

mason
mckenna
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sydney
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Michael
Ok Boomer
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ok boomer
sebastian
sydney
Alec S

Ashley
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Camden
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Ginaa
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Macon
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Ok Boomer
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Rhys
Shane
david

jason
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mckenna
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sydney
Alec S
Ashley
Bubble Blasters
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Ginaa

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Ok Boomer
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Rhys
Shane
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maddie
mason
mckenna

ok boomer
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sydney
Alec S
Ashley
Bubble Blasters
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Ginaa
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Lindsey
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Ok Boomer

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Ashley
Bubble Blasters

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Bubble Blasters
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ok boomer
sebastian

sydney
Alec S
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Bubble Blasters
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Ok Boomer
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Alec S
Ashley
Bubble Blasters
Camden
Chelsea

Dahlia
Ginaa
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Max
Michael
Ok Boomer
Owen
Rhys
Shane
david
jason
julia
kate
maddie

mason
mckenna
ok boomer
sebastian
sydney

Answer	Correct / Incorrect	Correct
A measure of the tendency of an atom to attract a bonding pair of electrons	Correct	1
A measure of the tendency of an atom to attract a bonding pair of electrons	Correct	1
	Incorrect	0
A measure of the tendency of an atom to attract a bonding pair of electrons	Correct	1
A measure of the tendency of an atom to attract a bonding pair of electrons	Correct	1
A measure of the tendency of an atom to attract a bonding pair of electrons	Correct	1
A measure of the tendency of an atom to attract a bonding pair of electrons	Correct	1
A measure of the tendency of an atom to attract a bonding pair of electrons	Correct	1
A measure of the tendency of an atom to attract a bonding pair of electrons	Correct	1
A measure of the tendency of an atom to attract a bonding pair of electrons	Correct	1
The attraction between an electron and the nucleus in any atom	Incorrect	0
A measure of the tendency of an atom to attract a bonding pair of electrons	Correct	1
	Incorrect	0
A measure of the tendency of an atom to attract a bonding pair of electrons	Correct	1
A measure of the tendency of an atom to attract a bonding pair of electrons	Correct	1

A measure of the tendency of an atom to attract a bonding pair of electrons	Correct	1
A measure of the tendency of an atom to attract a bonding pair of electrons	Correct	1
A measure of the tendency of an atom to attract a bonding pair of electrons	Correct	1
A measure of the tendency of an atom to attract a bonding pair of electrons	Correct	1
A measure of the tendency of an atom to attract a bonding pair of electrons	Correct	1
A measure of the tendency of an atom to attract a bonding pair of electrons	Correct	1
A measure of the tendency of an atom to attract a bonding pair of electrons	Correct	1
A measure of the tendency of an atom to attract a bonding pair of electrons	Correct	1
A measure of the tendency of an atom to attract a bonding pair of electrons	Correct	1
A measure of the tendency of an atom to attract a bonding pair of electrons	Correct	1
A measure of the tendency of an atom to attract a bonding pair of electrons	Correct	1
True	Correct	1
True	Correct	1
	Incorrect	0
True	Correct	1
True	Correct	1

RawReportData Data

	Incorrect	0
	Incorrect	0
True	Correct	1
True	Correct	1
True	Correct	1
True	Correct	1
True	Correct	1
	Incorrect	0
True	Correct	1
True	Correct	1
True	Correct	1
	Incorrect	0
True	Correct	1
True	Correct	1
True	Correct	1
False	Incorrect	0

RawReportData Data

True	Correct	1
True	Correct	1
True	Correct	1
True	Correct	1
True	Correct	1
Left to Right	Correct	1
Left to Right	Correct	1
	Incorrect	0
Left to Right	Correct	1
Down	Incorrect	0
Down	Incorrect	0
Left to Right	Correct	1
Left to Right	Correct	1
Left to Right	Correct	1
	Incorrect	0
Left to Right	Correct	1

RawReportData Data

Left to Right	Correct	1
	Incorrect	0
Left to Right	Correct	1
Left to Right	Correct	1
Left to Right	Correct	1
Left to Right	Correct	1
Left to Right	Correct	1
Left to Right	Correct	1
Left to Right	Correct	1
Down	Incorrect	0
Left to Right	Correct	1
Left to Right	Correct	1
Left to Right	Correct	1
Left to Right	Correct	1
Left to Right	Correct	1
Left to Right	Correct	1
Coulomb's Law	Correct	1

Coulomb's Law	Correct	1
	Incorrect	0
Coulomb's Law	Correct	1
Coulomb's Law	Correct	1
Coulomb's Law	Correct	1
Coulomb's Law	Correct	1
Coulomb's Law	Correct	1
Coulomb's Law	Correct	1
Coulomb's Law	Correct	1
Coulomb's Law	Correct	1
	Incorrect	0
Coulomb's Law	Correct	1
Coulomb's Law	Correct	1
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Coulomb's Law	Correct	1

Coulomb's Law	Correct	1
Coulomb's Law	Correct	1
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Coulomb's Law	Correct	1
Coulomb's Law	Correct	1
Coulomb's Law	Correct	1
Coulomb's Law	Correct	1
Coulomb's Law	Correct	1
Metals	Incorrect	0
Metals	Incorrect	0
	Incorrect	0
Nonmetals	Correct	1
Metals	Incorrect	0
Metals	Incorrect	0
Metals	Incorrect	0

RawReportData Data

Metals	Incorrect	0
Nonmetals	Correct	1
Nonmetals	Correct	1
Nonmetals	Correct	1
Metals	Incorrect	0
	Incorrect	0
Metals	Incorrect	0
Nonmetals	Correct	1
Nonmetals	Correct	1
Nonmetals	Correct	1
Nonmetals	Correct	1
Nonmetals	Correct	1
Nonmetals	Correct	1
Nonmetals	Correct	1
Nonmetals	Correct	1
Nonmetals	Correct	1

RawReportData Data

Nonmetals	Correct	1
Metals	Incorrect	0
Nonmetals	Correct	1
Ionic	Correct	1
Ionic	Correct	1
	Incorrect	0
Ionic	Correct	1
Ionic	Correct	1
Ionic	Correct	1
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Ionic	Correct	1
	Incorrect	0

Ionic	Correct	1
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Ionic	Correct	1
Ionic	Correct	1
When one atom transfers electrons to another	Correct	1
When one atom transfers electrons to another	Correct	1
	Incorrect	0

When one atom transfers electrons to another	Correct	1
A property that is measurable and whose value describes its state	Incorrect	0
When one atom transfers electrons to another	Correct	1
When one atom transfers electrons to another	Correct	1
When one atom transfers electrons to another	Correct	1
When one atom transfers electrons to another	Correct	1
When one atom transfers electrons to another	Correct	1
When one atom transfers electrons to another	Correct	1
When one atom transfers electrons to another	Correct	1
When one atom transfers electrons to another	Correct	1
	Incorrect	0
When one atom transfers electrons to another	Correct	1
When one atom transfers electrons to another	Correct	1
When one atom transfers electrons to another	Correct	1
When one atom transfers electrons to another	Correct	1
When one atom transfers electrons to another	Correct	1
When one atom transfers electrons to another	Correct	1

When one atom transfers electrons to another	Correct	1
A property that is measurable and whose value describes its state	Incorrect	0
When one atom transfers electrons to another	Correct	1
When one atom transfers electrons to another	Correct	1
When one atom transfers electrons to another	Correct	1
When one atom transfers electrons to another	Correct	1
When one atom transfers electrons to another	Correct	1
When two atoms share electrons	Correct	1
When two atoms share electrons	Correct	1
	Incorrect	0
When two atoms share electrons	Correct	1
When two atoms share electrons	Correct	1
	Incorrect	0
When two atoms share electrons	Correct	1
When two atoms share electrons	Correct	1
When two atoms share electrons	Correct	1

When two atoms share electrons	Correct	1
When two atoms share electrons	Correct	1
When two atoms share electrons	Correct	1
	Incorrect	0
When two atoms share electrons	Correct	1
	Incorrect	0
When two atoms share electrons	Correct	1
When two atoms share electrons	Correct	1
When two atoms share electrons	Correct	1
When two atoms share electrons	Correct	1
	Incorrect	0
Occurs when there is an electronegativity difference between bonded atoms	Incorrect	0
When two atoms share electrons	Correct	1
When two atoms share electrons	Correct	1
When two atoms share electrons	Correct	1
When two atoms share electrons	Correct	1

When two atoms share electrons	Correct	1
The distribution of electric charge around atoms or chemical groups	Correct	1
Elements with great differences in electronegativity	Incorrect	0
	Incorrect	0
Elements with great differences in electronegativity	Incorrect	0
Elements with great differences in electronegativity	Incorrect	0
	Incorrect	0
Elements with great differences in electronegativity	Incorrect	0
The distribution of electric charge around atoms or chemical groups	Correct	1
The distribution of electric charge around atoms or chemical groups	Correct	1
The distribution of electric charge around atoms or chemical groups	Correct	1
Elements with great differences in electronegativity	Incorrect	0
The distribution of electric charge around atoms or chemical groups	Correct	1
	Incorrect	0
The distribution of electric charge around atoms or chemical groups	Correct	1
The distribution of electric charge around atoms or chemical groups	Correct	1

Elements with great differences in electronegativity	Incorrect	0
The distribution of electric charge around atoms or chemical groups	Correct	1
Elements with great differences in electronegativity	Incorrect	0
The distribution of electric charge around atoms or chemical groups	Correct	1
The distribution of electric charge around atoms or chemical groups	Correct	1
The distribution of electric charge around atoms or chemical groups	Correct	1
The distribution of electric charge around atoms or chemical groups	Correct	1
The distribution of electric charge around atoms or chemical groups	Correct	1
The distribution of electric charge around atoms or chemical groups	Correct	1
The distribution of electric charge around atoms or chemical groups	Correct	1
The distribution of electric charge around atoms or chemical groups	Correct	1
False	Correct	1
False	Correct	1
	Incorrect	0
False	Correct	1
False	Correct	1

RawReportData Data

	Incorrect	0
False	Correct	1
False	Correct	1
False	Correct	1
False	Correct	1
False	Correct	1
False	Correct	1
	Incorrect	0
False	Correct	1
False	Correct	1
False	Correct	1
False	Correct	1
False	Correct	1
False	Correct	1
False	Correct	1
False	Correct	1
True	Incorrect	0

RawReportData Data

False	Correct	1
False	Correct	1
False	Correct	1
False	Correct	1
False	Correct	1

RawReportData Data

Incorrect	Score (points)	Score without Answer Streak Bonus (points)
0	924	924
0	966	966
1	0	0
0	966	966
0	981	981
0	887	887
0	931	931
0	907	907
0	984	984
0	960	960
1	0	0
0	973	973
1	0	0
0	908	908
0	938	938

RawReportData Data

0	964	964
0	906	906
0	987	987
0	943	943
0	933	933
0	967	967
0	986	986
0	985	985
0	955	955
0	946	946
0	948	948
0	1054	954
0	1062	962
1	0	0
0	1069	969
0	1093	993

RawReportData Data

1	0	0
1	0	0
0	1071	971
0	1092	992
0	1070	970
0	992	992
0	1087	987
1	0	0
0	1041	941
0	1091	991
0	1074	974
1	0	0
0	1094	994
0	1078	978
0	1058	958
1	0	0

RawReportData Data

0	1082	982
0	1085	985
0	1077	977
0	1093	993
0	1092	992
0	1183	983
0	1188	988
1	0	0
0	1192	992
1	0	0
1	0	0
0	931	931
0	1123	923
0	1191	991
1	0	0
0	1084	984

RawReportData Data

0	1173	973
1	0	0
0	1137	937
0	1140	940
0	1176	976
0	985	985
0	1191	991
0	1155	955
0	1134	934
1	0	0
0	1189	989
0	1196	996
0	1186	986
0	1184	984
0	1152	952
0	1280	980

RawReportData Data

0	1288	988
1	0	0
0	1292	992
0	988	988
0	986	986
0	1091	991
0	1269	969
0	1291	991
0	973	973
0	1188	988
0	1282	982
1	0	0
0	1287	987
0	1282	982
0	1284	984
0	1087	987

RawReportData Data

0	1293	993
0	1284	984
0	1251	951
0	987	987
0	1292	992
0	1290	990
0	1278	978
0	1283	983
0	1282	982
1	0	0
1	0	0
1	0	0
0	1364	964
1	0	0
1	0	0
1	0	0

RawReportData Data

1	0	0
0	1379	979
0	1072	972
0	1284	984
1	0	0
1	0	0
1	0	0
0	1339	939
0	1363	963
0	1175	975
0	1395	995
0	1375	975
0	1337	937
0	1076	976
0	1385	985
0	1383	983

RawReportData Data

0	1386	986
1	0	0
0	1388	988
0	979	979
0	987	987
1	0	0
0	1489	989
0	987	987
0	936	936
0	920	920
0	957	957
0	1483	983
0	1127	927
0	1362	962
0	984	984
1	0	0

RawReportData Data

0	983	983
0	1441	941
0	1464	964
0	1256	956
0	1463	963
0	1444	944
0	1432	932
0	1172	972
0	1481	981
0	1497	997
0	1488	988
0	964	964
0	1466	966
0	1002	902
0	1019	919
1	0	0

RawReportData Data

0	1441	941
1	0	0
0	1007	907
0	991	891
0	986	886
0	1497	997
0	1197	897
0	1481	981
0	1015	915
1	0	0
0	1024	924
0	1412	912
0	1431	931
0	1376	976
0	1481	981
0	1449	949

RawReportData Data

0	1404	904
1	0	0
0	1482	982
0	1496	996
0	1474	974
0	1011	911
0	1460	960
0	1166	966
0	1172	972
1	0	0
0	1483	983
0	987	987
1	0	0
0	1153	953
0	1159	959
0	1496	996

RawReportData Data

0	1347	947
0	1482	982
0	1177	977
1	0	0
0	1185	985
1	0	0
0	1474	974
0	1494	994
0	1493	993
0	1458	958
1	0	0
1	0	0
0	1482	982
0	1497	997
0	1489	989
0	1184	984

RawReportData Data

0	1488	988
0	1215	915
1	0	0
1	0	0
1	0	0
1	0	0
1	0	0
1	0	0
0	1254	954
0	1493	993
0	1447	947
1	0	0
0	1261	961
1	0	0
0	1287	987
0	982	982

RawReportData Data

1	0	0
0	1491	991
1	0	0
0	1435	935
0	972	972
0	950	950
0	1481	981
0	1494	994
0	1466	966
0	1287	987
0	1487	987
0	1382	982
0	986	986
1	0	0
0	987	987
0	993	993

RawReportData Data

1	0	0
0	989	989
0	1384	984
0	1494	994
0	1420	920
0	992	992
0	1381	981
1	0	0
0	1379	979
0	1039	939
0	979	979
0	1495	995
0	993	993
0	1472	972
0	987	887
1	0	0

RawReportData Data

0	1469	969
0	1495	995
0	1467	967
0	1396	996
0	1478	978

RawReportData Data

Current Total Score (points)	Answer Time (%)
924	15.22%
966	6.78%
0	100.00%
966	6.89%
981	3.78%
887	22.56%
931	13.89%
907	18.67%
984	3.22%
960	8.00%
0	2.33%
973	5.33%
0	100.00%
908	18.44%
938	12.33%

RawReportData Data

964	7.11%
906	18.78%
987	2.67%
943	11.44%
933	13.44%
967	6.56%
986	2.78%
985	3.00%
955	9.00%
946	10.78%
948	10.33%
1978	9.11%
2028	7.67%
0	100.00%
2035	6.11%
2074	1.44%

RawReportData Data

887	100.00%
931	100.00%
1978	5.78%
2076	1.56%
2030	6.00%
992	1.56%
2060	2.56%
0	100.00%
1949	11.89%
2029	1.89%
2038	5.22%
906	100.00%
2081	1.22%
2021	4.44%
1991	8.44%
967	4.89%

RawReportData Data

2068	3.67%
2070	3.00%
2032	4.56%
2039	1.44%
2040	1.67%
3161	3.44%
3216	2.33%
0	100.00%
3227	1.56%
2074	8.22%
887	28.33%
1862	13.89%
3101	15.44%
3267	1.89%
2030	100.00%
2076	3.22%

RawReportData Data

3233	5.44%
0	100.00%
3086	12.67%
3169	12.00%
3214	4.78%
1891	3.00%
3272	1.78%
3176	9.00%
3125	13.11%
967	6.89%
3257	2.22%
3266	0.78%
3218	2.78%
3223	3.11%
3192	9.56%
4441	4.00%

RawReportData Data

4504	2.44%
0	100.00%
4519	1.56%
3062	2.33%
1873	2.89%
2953	1.78%
4370	6.22%
4558	1.78%
3003	5.44%
3264	2.33%
4515	3.67%
0	100.00%
4373	2.67%
4451	3.56%
4498	3.11%
2978	2.67%

RawReportData Data

4565	1.44%
4460	3.22%
4376	9.89%
1954	2.67%
4549	1.56%
4556	2.00%
4496	4.44%
4506	3.44%
4474	3.67%
4441	3.78%
4504	3.00%
0	100.00%
5883	7.22%
3062	2.33%
1873	5.00%
2953	4.44%

RawReportData Data

4370	6.44%
5937	4.11%
4075	5.67%
4548	3.11%
4515	8.11%
0	100.00%
4373	15.33%
5790	12.22%
5861	7.44%
4153	5.00%
5960	1.00%
5835	5.00%
5713	12.56%
3030	4.89%
5934	3.00%
5939	3.33%

RawReportData Data

5882	2.78%
4506	4.56%
5862	2.44%
5420	4.11%
5491	2.67%
0	100.00%
7372	2.22%
4049	2.67%
2809	12.89%
3873	16.00%
5327	8.67%
7420	3.33%
5202	14.67%
5910	7.67%
5499	3.11%
0	100.00%

RawReportData Data

5356	3.44%
7231	11.78%
7325	7.11%
5409	8.89%
7423	7.44%
7279	11.22%
7145	13.67%
4202	5.56%
7415	3.78%
7436	0.56%
7370	2.44%
5470	7.11%
7328	6.89%
6422	19.67%
6510	16.22%
0	100.00%

RawReportData Data

8813	11.78%
4049	5.67%
3816	18.56%
4864	21.89%
6313	22.78%
8917	0.67%
6399	20.56%
7391	3.89%
6514	17.00%
0	100.00%
6380	15.22%
8643	17.67%
8756	13.78%
6785	4.89%
8904	3.89%
8728	10.22%

RawReportData Data

8549	19.11%
4202	12.89%
8897	3.56%
8932	0.78%
8844	5.11%
6481	17.89%
8788	8.00%
7588	6.78%
7682	5.56%
0	100.00%
10296	3.33%
5036	2.56%
3816	100.00%
6017	9.33%
7472	8.11%
10413	0.78%

RawReportData Data

7746	10.56%
8873	3.67%
7691	4.56%
0	100.00%
7565	3.00%
8643	100.00%
10230	5.22%
8279	1.11%
10397	1.44%
10186	8.33%
8549	100.00%
4202	7.00%
10379	3.56%
10429	0.56%
10333	2.22%
7665	3.22%

RawReportData Data

10276	2.44%
8803	17.00%
7682	18.78%
0	100.00%
10296	6.44%
5036	7.22%
3816	100.00%
6017	20.00%
8726	9.11%
11906	1.44%
9193	10.56%
8873	3.89%
8952	7.89%
0	100.00%
8852	2.67%
9625	3.56%

RawReportData Data

10230	8.89%
9770	1.78%
10397	2.33%
11621	13.00%
9521	5.56%
5152	10.00%
11860	3.89%
11923	1.22%
11799	6.78%
8952	2.67%
11763	2.67%
10185	3.67%
8668	2.89%
0	100.00%
11283	2.67%
6029	1.33%

RawReportData Data

3816	100.00%
7006	2.22%
10110	3.11%
13400	1.22%
10613	16.00%
9865	1.56%
10333	3.89%
0	100.00%
10231	4.11%
10664	12.22%
11209	4.11%
11265	1.00%
11390	1.33%
13093	5.56%
10508	22.67%
5152	6.33%

RawReportData Data

13329	6.11%
13418	1.00%
13266	6.56%
10348	0.78%
13241	4.33%

Answer Time (seconds)
13,7
6,1
90
6,2
3,4
20,3
12,5
16,8
2,9
7,2
2,1
4,8
90
16,6
11,1

6,4
16,9
2,4
10,3
12,1
5,9
2,5
2,7
8,1
9,7
9,3
8,2
6,9
90
5,5
1,3

90
90
5,2
1,4
5,4
1,4
2,3
90
10,7
1,7
4,7
90
1,1
4
7,6
4,4

3,3
2,7
4,1
1,3
1,5
3,1
2,1
90
1,4
7,4
25,5
12,5
13,9
1,7
90
2,9

4,9
90
11,4
10,8
4,3
2,7
1,6
8,1
11,8
6,2
2
0,7
2,5
2,8
8,6
3,6

2,2
90
1,4
2,1
2,6
1,6
5,6
1,6
4,9
2,1
3,3
90
2,4
3,2
2,8
2,4

1,3
2,9
8,9
2,4
1,4
1,8
4
3,1
3,3
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2,1
4,5
4

5,8
3,7
5,1
2,8
7,3
90
13,8
11
6,7
4,5
0,9
4,5
11,3
4,4
2,7
3

2,5
4,1
2,2
3,7
2,4
90
2
2,4
11,6
14,4
7,8
3
13,2
6,9
2,8
90

3,1
10,6
6,4
8
6,7
10,1
12,3
5
3,4
0,5
2,2
6,4
6,2
17,7
14,6
90

10,6
5,1
16,7
19,7
20,5
0,6
18,5
3,5
15,3
90
13,7
15,9
12,4
4,4
3,5
9,2

17,2
11,6
3,2
0,7
4,6
16,1
7,2
6,1
5
90
3
2,3
90
8,4
7,3
0,7

9,5
3,3
4,1
90
2,7
90
4,7
1
1,3
7,5
90
6,3
3,2
0,5
2
2,9

2,2
15,3
16,9
90
5,8
6,5
90
18
8,2
1,3
9,5
3,5
7,1
90
2,4
3,2

8
1,6
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9
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1,2

90
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5,5
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3,9